TEXAS WATER DEVELOPMENT BOARD

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REPORT 71

RECONNAISSANCE OF THE CHEMICAL QUALITY OF SURFACE WATERS OF THE COLORADO RIVER BASIN, TEXAS

Bу

Donald K. Leifeste and Myra W. Lansford

Prepared by the U.S. Geological Survey in cooperation with the Texas Water Development Board

March 1968

TEXAS WATER DEVELOPMENT BOARD

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RECONNAISSANCE OF THE CHEMICAL QUALITY

OF SURFACE WATERS OF

THE COLORADO RIVER BASIN, TEXAS

ABSTRACT

The natural runoff in most of the Colorado River basin is of good chemical quality and is suitable for most municipal, industrial, and agricultural purposes.

The kinds and quantities of minerals dissolved in surface water of the basin are related to the geology of the area and to rainfall and streamflow characteristics, but the quality of the water in the Colorado River below Lake J. B. Thomas is influenced also by oil-field brines.

Most of the tributary streams yield surface water averaging less than 250 ppm (parts per million) in dissolved-solids content, but the saline inflow in the upper basin keeps the average concentration in the main stem above 250 ppm throughout its length. The discharge weighted-average concentrations of the Colorado River near San Saba and at Wharton for the period 1958-65 are 295 ppm and 255 ppm, respectively.

Surface water of the basin generally ranges from moderately hard (61 to 120 ppm) to very hard (over 180 ppm). From Lake J. B. Thomas to the mouth of Pecan Bayou, the Colorado River and most of its tributaries contain very hard water. Downstream from Pecan Bayou the water of the basin is hard.

The chloride concentration in surface water of the basin ranges from less than 50 ppm to several thousand ppm. In the upper basin where brines are reaching the streams, chloride concentrations of several thousand ppm are common. Most of the remainder of the basin yields water averaging less than 50 ppm, but the saline inflow in the upper basin keeps the average concentration in the main stem above 50 ppm all the way to Austin. Higher concentrations are found in the South Concho River and in the headwater reaches of Pecan Bayou probably because of oil-field operations.

All the major water-supply reservoirs contain water of acceptable quality for most uses. The quality of the water that will be stored in Robert Lee Reservoir will depend on the success of the upstream salt-water alleviation program. Water available for storage at potential reservoir sites is of good quality; the dissolved-solids concentration is usually less than 350 ppm.



RECONNAISSANCE OF THE CHEMICAL QUALITY

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OF SURFACE WATERS OF

THE COLORADO RIVER BASIN, TEXAS

INTRODUCTION

The investigation of the chemical quality of the surface waters of the Colorado River basin, Texas, is part of a statewide reconnaissance. Each major river basin in the State is being studied, and reports presenting the results of the studies and summaries of available chemical-quality data are being prepared. River basins on which reports have been completed and the area covered by this report are shown in Figure 1.

The purpose of this report is to present the available information on the water quality of the Colorado River basin that will further the proper development, control, and use of the water resources of the area. In the study, the following items were considered: the nature and amounts of mineral constituents in solution; the geologic, hydrologic, and cultural influences that determine the water quality; the amount and probable source of the salt discharged by the streams; and the suitability of the water for domestic supply, industrial use, and irrigation.

A network of daily chemical-quality stations on principal streams in Texas is operated by the U.S. Geological Survey in cooperation with the Texas Water Development Board and with other Federal and local agencies. This network has been inadequate to inventory completely the chemical quality of the surface waters of the State. To supplement the information being obtained by the network, a cooperative statewide reconnaissance by the U.S. Geological Survey and the Texas Water Development Board was begun in September 1961. In this study, samples for chemical analyses were collected periodically at numerous sites throughout Texas so that some quality-of-water information would be available where water-development projects are likely to be built. These data aid in the delineation of areas having water-quality problems and in the identification of probable sources of pollution, thus indicating areas where more detailed investigations are needed.

During the period September 1961 to September 1965, water-quality data were collected for the principal streams, the major reservoirs, a number of potential reservoir sites, and many tributaries in the Colorado River basin.

Agencies that have cooperated in the collection of chemical-quality and streamflow data include the U.S. Army Corps of Engineers, the Brown County Water Improvement District No. 1, the Colorado River Municipal Water District, the Lower Colorado River Authority, the Texas Electric Service Company, the Texas State Department of Health, and the cities of Austin, Brady, and San Angelo.

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COLORADO RIVER DRAINAGE BASIN

General Description

As measured by length and drainage area, the Colorado is the largest river that is wholly in Texas. The basin extends into eastern New Mexico, but the part of the basin there does not contribute to the river flow. The Colorado River rises in north-central Dawson County near Lamesa, on the southern High Plains, and flows southeastward to Matagorda Bay, on the middle Gulf Coast. The Texas part of the basin, which includes all or part of 50 counties, is about 500 miles long and varies in width from 7 miles in southern Colorado County to 160 miles in the Brown-McCulloch County area. The average width is about 80 miles. The area of the basin in Texas is 39,890 square miles, or 15.2 percent of the area of Texas.

The elevation at the point of origin of the Colorado River is about 3,000 feet above mean sea level. The river flows through a rolling, generally prairie terrain to the vicinity of San Saba County where it enters the rugged Hill Country of Central Texas. It then flows through a series of canyons, crosses the Balcones Escarpment at Austin, and continues across the Coastal Plain to the Gulf.

The principal tributaries, in downstream order, are: The Concho River, Pecan Bayou, and the San Saba, Llano, and Pedernales Rivers. All except Pecan Bayou are spring-fed, perennial streams that begin in the Edwards Plateau.

The average annual precipitation ranges from a minimum of 13 inches in the upper part of the drainage area to a maximum of 43 inches at the mouth of the river. The average for the basin in Texas is 28 inches. Average monthly precipitation at three U.S. Weather Bureau stations and annual precipitation for the period 1931-65 at one station are shown in Figure 2.

Runoff is defined as that part of the precipitation appearing in surface streams, and is the same as streamflow unaffected by artificial storage or diversion (Langbein and Iseri, 1960, p. 17). Temperature, seasonal distribution of rainfall, storm intensity, infiltration rates, and types and density of vegetation affect the amount of runoff from a drainage basin.

The average annual runoff in the Colorado River basin ranges from a maximum of 6.6 inches (350 acre-feet per square mile) near the mouth of the river to less than 1.0 inch (53 acre-feet per square mile) west of an approximate north-south line through San Angelo. The runoff decreases more or less uniformly from east to west along with the decrease in rainfall. The runoff varies widely from year to year and between periods of wet and dry years.

Annual runoff, expressed as mean discharge in cubic feet per second and as inches per year, is shown in Figure 2 for the gaging station Colorado River at Columbus for the period 1940-65. The contributing drainage area at the station is 29,170 square miles. Runoff ranged from 0.44 to 3.86 inches per year and averaged 1.37 inches during the 26-year period of record.

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Population and Municipalities

The population of the Colorado River basin in 1965 was about 850,000, which was about 8 percent of the total population of the State. Less than one-fourth of the people in the basin live on farms. Austin is the largest city in the basin, with a 1965 population of about 240,000. San Angelo, Midland, and Odessa are other cities that have more than 50,000 inhabitants. Twelve other cities had 1965 populations of more than 5,000.

Agricultural and Industrial Development

The basin's economic base is oil production and agriculture. The western part of the basin has a heavy concentration of oil fields and petrochemical industries. Ranching and farming throughout the basin support a wool industry, cottonseed oil plants, cattle marketing operations, textile plants, creameries, and other industries. Miscellaneous light manufacturing includes aircraft and boat fabrication. The chief crops are cotton, wheat, grain sorghum, vegetables, and sugar beets. State and Federal offices, The University of Texas, tourism, and recreation on the Highland Lakes contribute substantially to the Austin area's economy.

Development of Surface-Water Resources

The Colorado River basin contributes about 6 percent of the State's total runoff (Figure 3). Runoff increases from west to east with the increase in rainfall, and the quantity of surface water available for development differs widely between the upper and lower ends of the basin. In the High Plains the only surface water available is the small quantity periodically salvageable from playas.

The Texas Water Development Board reported that 1,258,000 acre-feet of water was used in the Colorado River basin in 1960. Of this amount only 173,300 acre-feet was from surface-water sources. Municipal and industrial use of surface water was 83,400 acre-feet. Surface water supplements groundwater supplies for some cities and provides the total supply for others. Cities using surface water impounded in the Colorado River basin include Colorado City, Big Spring, Odessa, Snyder, Sweetwater, San Angelo, Brady, Coleman, Brownwood, and Austin.

In 1964, 89,900 acre-feet of surface water was used for irrigation. In the middle part of the basin about 40,000 acre-feet of surface water was diverted from the Colorado and its tributaries to irrigate cotton, peanuts, pastures, hay, and feed crops. In the coastal rice area 40,000 acre-feet of surface water was diverted for irrigation.

The Colorado River basin has 21 major reservoirs existing or under construction as of December 31, 1966, with capacities ranging from 8,640 to 1,950,000 acre-feet. Table 1 lists these reservoirs and gives their capacities and uses. Several of the reservoirs in the upper part of the basin were built by cities or by water districts to supply water for local municipal and industrial use. Twin Buttes Reservoir was constructed by the U.S. Bureau of Reclamation for flood control, conservation storage, recreation, and irrigation. San Angelo Reservoir was constructed by the U.S. Army Corps of Engineers

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Table 1.--Reservoirs with capacities of 5,000 acre-feet or more in the Colorado River basin

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Reservoir	Year operation began	Stream	a/Total storage capacity (acre-feet)	Owner	County	Use	
Lake J. B. Thomas	1952	Colorado River	203,600	Colorado River Municipal Water District	Borden, Scurry	M, I, R	
Lake Colorado City	1949	Morgan Creek	31,600	Texas Electric Service Company	Mitchell	м, і	
Champion Creek	1959	Champion Creek	42,500	do	do	м, і	
Robert Lee	*	Colorado Ríver	520,000	Colorado River Municipal Water District	Coke	м, і	
Oak Creek	1952	Oak Creek	39,360	City of Sweetwater	do	м, і	
Twin Buttes	1962	Middle and South Concho Rivers	640,600	U.S. Government	Tom Green	M, I, Ir, R, FC	
Lake Nasworthy	1948	South Concho River	12,390	City of San Angelo	do	M, I, Ir, R	
San Angelo	1952	North Concho River	396,400	U.S. Government	do	M, I, Ir, R, FC	
Hords Creek	1948	Hords Creek	8,640	do	Coleman	M, I, FC	
Coleman	1966	Jim Ned Creek	40,000	do	do	м, і	
Brownwood	1933	Pecan Bayou	143,400	Brown County Water Improvement District No. l	Brown	M, I, Ir	
Brady Creek	1963	Brady Creek	30,430	City of Brady	McCulloch	м, і	
Lake Buchanan	1938	Colorado River	992,000	Lower Colorado River Authority	Llano, Burnet	M, I, Ir, P, R	
Inks Lake	1938	do	17,500	do	do	P, R	
Lake Lyndon B. Johnson	1951	do	145,200	do	do	P, R	
Marble Falls Lake	1951	do	8,760	do	do	P, R	
Lake Travis	1942	do	1,950,000	do	Travis	M, I, Ir, R, P, FC	
Lake Austin	1939	do	21,000	do	do	M, R, P	
Decker Lake	*	Decker Greek	33,940	City of Austin	do	I, R	
Lake Bastrop	1964	Spicer Creek	16,590	Lower Colorado River Authority	Bastrop	Ir	
Eagle Lake	1900	Colorado River (off channel)	9,600	Lakeside Irrigation Company	Colorado	Ir	

The purpose for which the impounded water is used is indicated by the following symbols: M, municipal; I, industrial; Ir, irrigation; Mi, mining; R, recreation; P, hydroelectric power; FC, flood control

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* Under construction as of December 31, 1966. a Total capacity is that capacity below the lowest uncontrolled outlet or spillway (in some cases top of gates) and is based on the most recent reservoir survey available.

primarily for flood control and municipal supply. The six Lower Colorado River Authority lakes--Buchanan, Inks, Lyndon B. Johnson, Marble Falls, Travis, and Austin--are operated as a unit for generating hydroelectric power. Buchanan provides conservation storage, and Travis provides both conservation storage and flood control. Lake Bastrop, owned by the Lower Colorado River Authority, provides cooling water for a steam-electric generating plant.

Projects under construction in the Colorado River basin include Robert Lee Reservoir for additional municipal and industrial supply for the cities of Big Spring, Odessa, Snyder, and Midland; and Decker Lake to provide cooling water for a new steam-electric generating plant near Austin.

Figure 4 shows locations of the existing reservoirs, the two reservoirs under construction, and a number of potential dam sites which have been considered by various agencies.

The Soil Conservation Service of the U.S. Department of Agriculture was authorized by the Flood Control Act of 1936 to investigate and prescribe measures for runoff and water-flow retardation and soil-erosion prevention. As of September 30, 1966, 207 upstream floodwater-retarding structures had been built under this program in the Colorado River basin. These structures partly control flow from 1,180 square miles. Nineteen of the reservoirs are in the Cummings Creek subbasin in Fayette and Lee Counties in the lower part of the basin. The remaining 188 structures are in Callahan, Coke, Runnels, Menard, Schleicher, McCulloch, Coleman, and Brown Counties in the northwestern and north-central part of the basin.

CHEMICAL QUALITY OF THE WATER

Chemical-Quality Records

The U.S. Geological Survey has been collecting quality-of-water data in the Colorado River basin since 1944 when a daily-sampling station was established on the Colorado River at Wharton. Currently (1966), eleven dailysampling stations are in operation. In addition to collecting daily samples for chemical analyses, the Geological Survey has operated a continuously recording conductivity meter on the Colorado River near Cuthbert since March 1965.

Collection of chemical-quality data for this reconnaissance began in 1961 and continued through September 1965. Samples were collected periodically from most of the principal tributary streams and reservoirs. Numerous miscellaneous samples have been collected by the Geological Survey since 1941, and the results of the analyses of these samples have been included in this report. Most of the sampling for this study was done at gaging stations. When sampling was done at other sites, discharge measurements were usually made when the samples were collected.

The periods of record of all data-collection sites are shown on Table 4 and the locations are shown on Figure 10. The chemical-quality data for the daily stations are summarized in Table 5, and the complete records are published in an annual series of U.S. Geological Survey Water-Supply Papers and in reports of the Texas Water Development Board. (See table in the list of references.) Results of all the periodic and miscellaneous analyses are given in Table 6. The Texas State Department of Health makes available to the U.S. Geological Survey the data collected in its statewide stream-sampling program, which includes the periodic determination of pH, biochemical oxygen demand, total solids, dissolved oxygen, chloride, chlorine demand, and sulfate at 26 locations in the Colorado River basin. The data-collection sites of the Texas State Department of Health are listed in the following table. Most of them are at Geological Survey gaging stations and the numbers refer to locations on Figure 10.

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Location No.	State Department of Health data-collection site
9	Colorado River at Colorado City
	Beals Creek at FR 821 near Big Spring
25	Colorado River at Robert Lee
30	Colorado River at Ballinger
36	South Concho River at Christoval
38	Middle Concho River near Tankersley
53	North Concho River near Carlsbad
58	Concho River near San Angelo
59	Concho River near Paint Rock
62	Colorado River at Winchell
80	Pecan Bayou at Brownwood
84	San Saba River at Menard
89	San Saba River at San Saba
91	Colorado River near San Saba
	Colorado River near San Saba
96	Llano River near Junction
99	Llano River at Llano
103	Pedernales River near Johnson City
	Colorado River below Mansfield Dam
114	Colorado River at Austin
122	Colorado River at Bastrop

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Location No.	State Department of Health data-collection site
125	Colorado River at Smithville
130	Colorado River at Columbus
133	Colorado River at Wharton
134	Colorado River near Bay City
	Colorado River at Matagorda above Intracoastal Canal

Streamflow Records

Streamflow records in the Colorado River basin date from 1894, when the U.S. Geological Survey began collecting gage-height records of the Colorado River at the dam above Austin. The daily flow of the Colorado River at Austin has been measured continuously since 1898. More than 40 years of continuous discharge records are available for several stations on the main stem of the Colorado River, and records for more than 20 years are available for many of the principal tributaries. In 1966 the Geological Survey operated 57 streamflow stations, 12 reservoir-content stations, 13 low-flow partial-record stations, and 19 crest-stage partial-record stations. During this reconnais-sance, discharge measurements were made at other sites where water samples were collected for chemical analysis.

The periods of record for all the streamflow stations are given in Table 4 and the locations are shown on Figure 10. Records of discharge and stage of streams and contents and stages of lakes or reservoirs from 1898 to 1960 have been published in the annual series of U.S. Geological Survey Water-Supply Papers. (See table in the list of references.) Beginning with the 1961 water year, streamflow records have been released by the Geological Survey in annual state reports (U.S. Geological Survey, 1961, 1962, 1963, 1964b, 1965, 1966). Summaries of discharge records giving monthly and annual totals have been published (Texas Board of Water Engineers, 1958; U.S. Geological Survey, 1960, 1964a).

Environmental Factors and Their Effects on the Chemical Quality of the Water

Many environmental factors determine the chemical quality of a water, the most important of which are geology, patterns and characteristics of streamflow, and the activities of man.

Geology

When industrial and municipal influences are small, the chemical character of a river water is dependent primarily upon the composition of the geologic formations that are traversed and the time that the water is in contact with the rocks.

The amount of soluble minerals in rocks and soils is decreased by leaching. In arid or semiarid regions, most soils and the rocks from which they originated are incompletely leached and still contain large quantities of readily soluble material. Conversely, in areas of high rainfall, the mantle rock and residual soil contain relatively small amounts of readily soluble minerals. In the Colorado River basin, where the average annual precipitation varies from less than 13 inches in the northwestern part to over 42 inches near the coast, the amount of leaching varies geographically. Partly because of this, the dissolved-solids content of surface runoff and of ground-water inflow to streams is greatest in the western part of the basin and decreases toward the coast.

Figure 5 shows the geochemical character and ionic concentration of some surface waters in the Colorado River basin. The total ionic concentration in equivalents per million is equal to twice the length of either the vertical or horizontal axis. If the major part of the quadrilateral is in the lower left quarter, sulfate or chloride predominate among the anions, and sodium or potassium among the cations. If the major part is in the upper right quarter, calcium or magnesium, and carbonate or bicarbonate, are predominant.

Headwaters of the Colorado River rise primarily on the Dockum Group of Triassic age, which is composed of sand and shale. Water from this area, represented by the water stored in Lake J. B. Thomas, is generally dilute, of a mixed type, and has sodium and bicarbonate as its principal ions. Downstream from Lake J. B. Thomas, the Colorado River traverses sediments of Permian age composed of sand, shale, limestone, anhydrite, and salt. This is an area of saline inflow that degrades much of the water of the upper basin.

Inflow from small tributaries dilutes the water in the main stem by the time it reaches Ballinger, but sodium and chloride are still the predominant ions. Salinity is further decreased between the chemical-quality stations at Ballinger and near San Saba due to inflow from three major tributaries. Two of these, the Concho and San Saba Rivers, rise on Cretaceous rocks composed mainly of limestone, shale, sand, and silt. Runoff from these formations is generally of the calcium bicarbonate type and is dilute. Pecan Bayou primarily drains rocks of Pennsylvanian age, which are composed of marine sand, shale, and limestone. The water contributed by Pecan Bayou is of a mixed type and is low in dissolved solids. The inflow from these tributaries dilutes the main stem to a dissolved-solids content less than 300 ppm (parts per million); calcium and bicarbonate are the predominant ions.

Downstream from San Saba the tributaries that influence the chemical quality of the Colorado River are the Llano and Pedernales Rivers, which enter the main stem in the chain of Highland Lakes above Austin. Both of these streams drain a limestone terrain and contribute water saturated or nearly saturated with calcium and bicarbonate.

No major tributaries enter the Colorado River downstream from Austin, and most of the flow in the river is maintained by releases from the Highland Lakes. The quality of the water is uniform from Austin to the mouth. The water usually contains about 250 ppm of dissolved solids and is calcium bicarbonate in type.

Streamflow [Variable]

The patterns and characteristics of streamflow usually affect the chemical character of water in streams. In most streams where the flow is not regulated by upstream reservoirs, the concentrations of dissolved-mineral constituents vary inversely with the flow of the stream. The base flow, or sustained low

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flow, of a stream is predominantly water that has entered the stream from the ground-water reservoir. Usually this water has been in contact with rocks and soils for a sufficient time to dissolve part of their soluble minerals. At high stages most of the flow of a stream consists of surface runoff that has been in contact with rocks and soils for only a short time. Therefore, the dissolved-solids concentration of the stream is usually lowest during periods of high flow. This relationship is applicable in the upper Colorado River basin, but in the central portion of the basin where streams drain a limestone terrain, dissolved-solids concentrations vary only slightly with changes in discharge. In the lower part of the basin, streamflow is sustained by releases from the Highland Lakes, and tributary inflow is not sufficient to affect greatly the quality of the main stem.

Although the dissolved-solids concentration of the upper Colorado River is related in a general way to water discharge, the dissolved-solids concentration of the water cannot be estimated from streamflow data. The first increased streamflow resulting from a particular rain is usually more saline than an equal discharge that occurs later. Tributary inflow may also contribute significantly to streamflow but not be dilute enough to affect the concentration of the flow in the main stem. Figure 6 is a plot of electrical conductance of the water and discharge of the Colorado River near Cuthbert. Conductance is a measure of the total concentration of ions in water and can be directly related to dissolved-solids content. The general inverse relation of discharge to concentration of dissolved solids is well shown in Figure 6A, which shows a short duration rise following a period of no flow. Figure 6B shows a short duration rise following a period of low flow. The initial decrease in conductance is caused by local runoff; the sharp increase in conductance while the discharge was still increasing is the flushing out effect of the runoff from upstream. Figure 6C is a plot of an extended runoff event. The flushing out effect is again obvious. Discharge varies considerably during this rise but the conductance remains fairly uniform after the first 24 hours.

Duration curves of dissolved solids and water discharge for the Colorado River near San Saba and Colorado River at Wharton are given in Figure 7. The dissolved solids duration curve is a cumulative frequency curve that shows the percent of time during which specified dissolved-solids concentrations were equaled or exceeded during a specified period. The flow-duration curve is a cumulative frequency curve that shows the percent of time that water discharge was equal to or less than a specified discharge. The curves, therefore, show the inverse relation of dissolved-solids concentration to water discharge. For example, the Colorado River near San Saba had a dissolved-solids concentration of more than 285 ppm and flowed at a rate of less than 640 cfs (cubic feet per second) 80 percent of the time, and had a dissolved-solids concentration of more than 750 ppm and a discharge of less than 18 cfs 5 percent of the time, during the period 1948-65.

Activities of Man

The activities of man often have a deteriorative effect on the chemical quality of water. Oil-field brine, municipal and industrial wastes, and irrigation return flows increase the concentration of dissolved materials in streams. Evaporation from reservoirs increases the dissolved-solids concentration of the remaining water.



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Oil is produced in many areas in the Colorado River basin (Figure 8). Brine is produced in nearly all oil fields and it may, if improperly handled, eventually reach the streams. The composition of oil-field brine varies; but the principal chemical constituents in order of magnitude of their concentration (in ppm) are generally chloride, sodium, calcium, and sulfate. The quality of the water in the Colorado River and Beals Creek in Mitchell, Howard, and Scurry Counties is seriously affected by brines. Investigators in the past have disagreed as to the origin of the brine, but Reed (1961) in a consulting report to the Colorado River Municipal Water District presents convincing evidence that the brines entering the river are directly related to oil-field operations. The purpose of Reed's study was to determine the various sources of salt water present in the Colorado River principally in the area between Lake J. B. Thomas and Colorado City, a distance of about 24 river miles. The study was divided into three parts: first, a detailed study of the geology of the upper 1,000 feet of section with particular emphasis on the nature and structure of the surface beds which provide all the low flow of the Colorado River; second, a study of the ground water adjacent to the Colorado River and its tributaries, including chemical analyses and determinations of the altitude of the water table; and third, a study of U.S. Geological Survey quality and flow data of the Colorado River together with measurements of the thickness of underflow gravel in the river channel. As a result of his study, Reed concluded:

- 1. The probable maximum chloride ion concentration of the Colorado River prior to the development of the oil fields between the present site of Lake J. B. Thomas and Colorado City, Texas, was of the order of 300 to 500 parts per million during periods of maximum evaporation.
- 2. There is no known source of <u>natural</u> inflow of salt water to the river with chlorides significantly higher than 500 parts per million.
- 3. A great percentage of the total mineral content of brines produced with oil in the watershed of the Colorado River does eventually find its way into the Colorado River itself.
- 4. There are an unknown number of abandoned oil wells which were improperly plugged and which are now contributing salt water to the Colorado River and which must be controlled.

An indication of the magnitude of the man-made pollution problem is given by the tremendous volume of salt water produced with oil or gas in the area. The Texas Water Commission and Texas Water Pollution Control Board (1963) compiled an inventory conducted by the Railroad Commission of Texas which showed that approximately 222,400,000 barrels (28,680 acre-feet) of brine was produced in 1961 in the Colorado River basin in Texas. Of this amount, 63.2 percent or 140,650,000 barrels (18,130 acre-feet) was reinjected into the subsurface, with the remaining 10,550 acre-feet being placed in unlined surface pits or dumped directly into surface watercourses. Some of the salt water reinjected into the subsurface also contributed to the problem because of inadequately completed injection wells and improperly plugged abandoned wells and test holes. Robert Lee Reservoir is being built (1966) on the Colorado River downstream from the area of saline inflow. The Colorado River Municipal Water District plans to divert the low flows of the river upstream from the reservoir, and impound only the storm runoff, thereby allowing only the better-quality water to enter the reservoir. Extensive cleanup measures in the oil fields will also be necessary to insure that the water impounded will be satisfactory for municipal use.

Oil-field brines are also causing some deterioration of water quality in the Concho River and Pecan Bayou subbasins.

Municipal use of water tends to increase the concentration of dissolved solids in a stream system. The depletion of flow by diversion and consumptive use, the loss of water because of increased evaporation, and the disposal of municipal wastes into a stream result in higher average concentrations of dissolved solids in the remaining water. The municipal use of water in the Colorado River basin has not caused significant changes in water quality, but the disposal of municipal and industrial wastes has degraded the quality of the water of the Colorado River immediately downstream from Austin.

The waste load carried by a stream can also have significant effects on the water impounded in downstream reservoirs. Connell (1964) was especially concerned about the increasing phosphate in Texas reservoirs and the potential phosphate loading of many of the projected reservoirs. He lists the principal source of phosphate as municipal and industrial waste water, but also says surface runoff may contribute significantly to the phosphate content of streams and reservoirs, from leaching and erosion of mineral phosphate from soil, decay of vegetation and animal wastes, and use of phosphate fertilizers and phosphorus-containing insecticides. Dr. Connell lists the following serious quality threats caused by phosphate loading of projected reservoirs:

> First, production of excessive biological activity and associated odors and tastes rendering the water difficult and expensive to purify for domestic use. Second, promotion of heavy algal bloom and subsequent oxygenconsuming decay of organic matter sufficient to kill fish and to render water undesirable for recreation activities. Third, production of sufficient organic matter--"slimy soupy growth"--to render water difficult and expensive to process, distribute and use for industrial purposes. Fourth, very objectionable calcium phosphate scaling in cooling and boiler water uses.

Algal blooms may be promoted by as small amounts as 0.05 to 0.1 ppm of inorganic phosphate, or 0.2 to 0.6 ppm of inorganic plus organic phosphate (phosphate expressed in equivalent PO_4). Other factors favoring development of algal bloom are presence of essential nutrients, quiescence, clear water, and abundant sunlight. These conditions will frequently be attained in many reservoirs.

Connell (1966) reports phosphate concentrations as high as 0.3 to 0.4 ppm in the Colorado River at Wharton and Bay City. He concludes that 90 percent of this total is attributable to the municipal and industrial waste water from the city of Austin. In reference to the proposed Columbus Bend Reservoir he makes the following statement.

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The periodic production and deposition of organic matter through algal growth, and subsequent lifting and transport, have not been of sufficient proportions to produce seriously adverse effects on the current use of the water, i.e. for fishing, boating, and irrigation. But some curtailment of phosphate sources will probably be necessary for adequate protection of the quality of water in Columbus Bend Reservoir.

Relation of Quality of Water to Use

Quality-of-water studies are usually concerned with determining the suitability of water--judged by the chemical, physical, and biological characteristics--for its proposed use. In the Colorado River basin, surface water is used primarily for municipal and industrial supplies and for irrigation. This report considers only the chemical character of the water and its relationship to the principal uses.

All natural water contains dissolved-mineral matter. Most of this mineral matter in water is dissociated into charged particles, or ions. Principal cations (positive-charged ions) in natural water are calcium (Ca), magnesium (Mg), sodium (Na), potassium (K), and iron (Fe). The principal anions (negative-charged ions) are carbonate (CO₃), bicarbonate (HCO₃), sulfate (SO₄), chloride (C1), fluoride (F), and nitrate (NO₃). Other constituents and properties are often determined to help define the chemical and physical quality of water. Table 2 lists the constituents and properties commonly determined by the U.S. Geological Survey, and includes a resume of their sources and significance.

Domestic Supply

Because of differences in individuals, varying amounts of water used, and other factors, the safe limits for mineral constituents in drinking water are difficult to define. The limits usually accepted in the United States are the drinking-water standards established by the United States Public Health Service. Originally established in 1914 to control the quality of water used on interstate carriers for drinking and culinary purposes, these standards have been revised several times. The latest revision was in 1962 (U.S. Public Health Service, 1962). These standards have been accepted by the American Water Works Association and by most of the state departments of public health as minimum standards for all public water supplies.

The maximum concentrations permitted by these standards are given for selected constituents in the table on page 28.

Table	2.	Source	and	significance	of	dissolved	mineral	constituents	and	properties	of	water
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Constituent or property	Source or cause	Significance
Silica (SiO ₂)	Dissolved from practically all rocks and soils, commonly less than 30 ppm. High concentrations, as much as 100 ppm, generally occur in highly alka- line waters.	Forms hard scale in pipes and boilers. Carried over in steam of high-pressure boilers to form deposits on blades of turbines. Inhibits deterioration of zeolite-type water softeners.
Iron (Fe)	Dissolved from practically all rocks and soils. May also be derived from iron pipes, pumps, and other equip- ment. More than l or 2 ppm of iron in surface waters generally indicate acid wastes from mine drainage or other sources.	On exposure to air, iron in ground water oxidizes to reddish-brown precipitate. More than about 0.3 ppm stain laundry and utensils reddish-brown. Objectionable for food processing, textile proces- sing, beverages, ice manufacture, brewing, and other processes. U.S. Public Health Service (1962) drinking-water standards state that iron should not exceed 0.3 ppm. Larger quantities cause unpleasant taste and favor growth of iron bacteria.
Calcium (Ca) and Magnesium (Mg)	Dissolved from practically all soils and rocks, but especially from lime- stone, dolomite, and gypsum. Calcium and magnesium are found in large quantities in some brines. Magnesium is present in large quantities in sea water.	Cause most of the hardness and scale-forming properties of water; soap consuming (see hardness). Waters low in calcium and magnesium desired in electroplating, tanning, dyeing, and textile manufac- turing.
Sodium (Na) and Potassium (K)	Dissolved from practically all rocks and soils. Found also in ancient brines, sea water, industrial brines, and sewage.	Large amounts, in combination with chloride, give a salty taste. Moderate quantities have little effect on the usefulness of water for most purposes. Sodium salts may cause foaming in steam boilers and a high sodium content may limit the use of water for irrigation.
Bicarbonate (HCO3) and Carbonate (CO3)	Action of carbon dioxide in water on carbonate rocks, such as limestone and dolomite.	Bicarbonate and carbonate produce alkalinity. Bicarbonates of cal- cium and magnesium decompose in steam boilers and hot-water facili- ties to form scale and release corrosive carbon dioxide gas. In combination with calcium and magnesium, cause carbonate hardness.
Sulfate (SO4)	Dissolved from rocks and soils con- taining gypsum, iron sulfides, and other sulfur compounds. Commonly pre- sent in mine waters and in some indus- trial wastes.	Sulfate in water containing calcium forms hard scale in steam boilers. In large amounts, sulfate in combination with other ions gives bitter taste to water. Some calcium sulfate is considered beneficial in the brewing process. U.S. Public Health Service (1962) drinking-water standards recommend that the sulfate content should not exceed 250 ppm.
Chloride (Cl)	Dissolved from rocks and soils. Pre- sent in sewage and found in large amounts in ancient brines, sea water, and industrial brines.	In large amounts in combination with sodium, gives salty taste to drinking water. In large quantities, increases the corrosiveness of water. U.S. Public Health Service (1962) drinking-water stan- dards recommend that the chloride content should not exceed 250 ppm.
Fluoride (F)	Dissolved in small to minute quanti- ties from most rocks and soils. Added to many waters by fluoridation of municipal supplies.	Fluoride in drinking water reduces the incidence of tooth decay when the water is consumed during the period of enamel calcifica- tion. However, it may cause mottling of the teeth, depending on the concentration of fluoride, the age of the child, amount of drinking water consumed, and susceptibility of the individual. (Maier, F. J., 1950.)
Nitrate (NO ₃)	Decaying organic matter, sewage, fertilizers, and nitrates in soil.	Concentration much greater than the local average may suggest pol- lution, U.S. Public Health Service (1962) drinking-water standards suggest a limit of 45 ppm. Waters of high nitrate content have been reported to be the cause of methemoglobinemia (an often fatal disease in infants) and therefore should not be used in infant feeding. Nitrate has been shown to be helpful in reducing inter- crystalline cracking of boiler steel. It encourages growth of algae and other organisms which produce undersirable tastes and odors.
Dissolved solids	Chiefly mineral constituents dis- solved from rocks and soils. Includes some water of crystallization.	U.S. Public Health Service (1962) drinking-water standards recommend that waters containing more than 500 ppm dissolved solids not be used if other less mineralized supplies are available. Waters containing more than 1000 ppm dissolved solids are unsuita- ble for many purposes.
Hardness as CaCO ₃	In most waters nearly all the hardness is due to calcium and magnesium. All the metallic cations other than the alkali metals also cause hardness.	Consumes soap before a lather will form. Deposits soap curd on bathtubs. Hard water forms scale in boilers, water heaters, and pipes. Hardness equivalent to the bicarbonate and carbonate is called carbonate hardness. Any hardness in excess of this is called noncarbonate hardness. Waters of hardness as much as 60 ppm are considered soft; 61 to 120 ppm, moderately hard; 121 to 180 ppm, hard; more than 180 ppm, very hard.
Specific conductance (micromhos at 25° C)	Mineral content of the water.	Indicates degree of mineralization. Specific conductance is a mea- sure of the capacity of the water to conduct an electric current. Varies with concentration and degree of ionization of the consti- tuents.
Hydrogen ion concentration (pH)	Acids, acid-generating salts, and free carbon dioxide lower the pH. Carbo- nates, bicarbonates, hydroxides, and phosphates, silicates, and borates raise the pH.	A pH of 7.0 indicates neutrality of a solution. Values higher than 7.0 denote increasing alkalinity; values lower than 7.0 indicate increasing acidity. pH is a measure of the activity of the hydro- gen ions. Corrosiveness of water generally increases with decreas- ing pH. However, excessively alkaline waters may also attack metals.

	Maximum
Constituents	concentration (ppm)
Sulfate	250
Chloride	250
Nitrate	45
Fluoride	<u>a</u> / .9
Dissolved solids	500

a/ Based on temperature records for Austin.

In the Colorado River basin the concentrations of these constituents are generally lower than the maximum concentrations recommended by the U.S. Public Health Service. The exception is the area between Ira and Ballinger in the upper Colorado River basin, where concentrations frequently exceed these recommended limits.

Industrial Use

The industrial use of water in the Colorado River basin is primarily as cooling water for steam generators and for the generation of hydroelectric power, but additional industrial development in the basin is expected and surface water will probably be used to meet the demands. The quality requirements vary greatly for almost every industrial application, as indicated by the waterquality tolerances given in Table 3. One requirement of most industries is that concentrations of various constituents of the water remain relatively constant. When concentrations of undesirable substances in water vary, constant monitoring is required, and thus operating expenses are increased.

Hardness is one of the more important properties of water that affects its utility for industrial purposes. Excessive hardness is objectionable because it contributes to the formation of scale in steam boilers, pipes, water heaters, and various other equipment where water is heated, evaporated, or treated with alkaline materials. The accumulation of scale increases costs for fuel, labor, repairs and replacement, and lowers the quality of many wet-processed products. However, some calcium hardness may be desirable because calcium carbonate sometimes forms protective coatings on pipes and other equipment and reduces corrosion.

The corrosive property of water receives considerable attention in industrial water supplies. A high concentration of dissolved solids in a water may be closely associated with the corrosive property of the water, especially if chloride is present in appreciable quantities. Water that contains a large concentration of magnesium chloride may be highly corrosive because the hydrolysis of this salt yields hydrochloric acid.

The surface water of the Colorado River basin is hard and often sufficiently mineralized to require treatment for many industrial uses. However, it usually is satisfactory as cooling water. Table 3.--Water-quality tolerances for industrial applications \mathcal{Y}

[Allowable limits in parts per million except as indicated]

Gen-eral 2

Na2SO4 to Na2SO3 ratio

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Mn	0.5	:::	-: -:	5.5	4444	ci ci	.02	. 1. 5	.05	50.3	.25 .25	.2	
е И	0.5	:::			4494	~~~	.02	1.0 .1	г.	20.05	.25	.2	
Ca	11	:::	100-200 200-500	::	::::	::	:	:::	1	111	: : :	:	
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Ηd	::	8.04 0.54 0.4	6.5-7.0 7.0→	::	1911	::	;	; ; ;	;	7.8-8.3 8.0		;	
Alka- linity (as CaCO ₃)	::	:::	75 150	::	2 1 1 1 2	30-50	;	:::	;	50 	:::	;	
Hard- ness	(4)	75 40 8	11	25-75	250 	50	;	180 100 100	50	8 55 50-135	500 500	20	
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Lndustry	Air conditioning <u>3</u> / Baking	Boiler feed: 0-150 psi 150-250 psi 250 psi and up	Brewing: 5/ Light Dark	Canning: Legumes General	Carbonated bev- crages <u>9</u> Confectionary Cooling <u>9</u> Food, general	Ice (raw water) <u>9</u> Laundering	riasrics, clear, undercolored	Paper and pulp: 10/ Groundwood Kraft pulp Soda and sulfite	HL-Grade	Rayon (viscose) pulp: Production Manufacture Tanning <u>L</u> J	Textiles: General Dyeing <u>12</u> Wool scouring <u>13</u> Corron hand-	age 13/	<u> M</u> American Water

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2 A --Ho correstences is --to since formation; C--Conformance to Federal drinking-water standards necessary; D--NaCl, 275 ppm. 5 Marcers with signe and hydrogen suffice odors are most unsuitable for air conditioning. 5 Some hordness desained formation; C--Conformance to Federal drinking-water standards necessary; D--NaCl, 275 ppm. 5 Some hordness desained formation; C--Conformance to Federal drinking-water standards necessary; D--NaCl, 275 ppm. 5 Some hordness desained formation; C--Conformance to Federal drinking-water of light-beer quality; whiskey mashing water of dark-beer quality). 5 Mater for distilling must meet the same general requirements as for brewing (gin and spirits mashing water of light-beer quality; whiskey mashing water of dark-beer quality). 5 Mater dorbless, starlie water for syrup and eurobrization. Water consistent in character: Nake quality filtered municipal water not satisfactory for beverages. 7 Mater dorbless is necessary as low value favors investion of success, cassing sticky product. 8 Control of correstences is necessary as is also control of organisms, such as sulfur and iron batteria, which tend to form since. 9 Ga(E003) particulary tronblessone - Mg(E003)2 tends to prevent cracking. Sulfates and chlorides of Ga, Mg, Na should each be less than 300 ppm (white butts). 9 Ga(E003) particulary tronblessone - Mg(E003)2 tends to prevent cracking. Sulfates and chlorides of Ga, Mg, Na should each be less than 300 ppm (white butts). 9 Ga(E003) particulary tronblessone - Mg(E003)2 tends to prevent cracking. Sulfates and chlorides of Ga, Mg, Na should each be less than 300 ppm (white butts). 9 Ga(E003) particulary tronblessone - Mg(E003)2 tends to prevent cracking. Sulfates and chlorides of Ga, Mg, Na should each be less than 300 ppm (white butts). 9 Ga(E003) particulary tronblessone - Mg(E003)2 tends to prevent cracking. Sulfates and chlorides of Ga, Mg, Ma should each be less than 300 ppm (for beverse by chlorine, causing reddish color. 9 Ga(E003) particulary tronblessone - Mg(E003)2

Irrigation

The extent to which chemical quality limits the suitability of a water for irrigation depends on such factors as: the nature, composition, and drainage of the soil and subsoil; the amounts of water used and the methods of applying it; the kind of crops grown; and the climate of the region. Because these factors are highly variable, every method of classifying waters for irrigation is somewhat arbitrary.

The most important characteristics in determining the quality of irrigation water, according to the U.S. Salinity Laboratory Staff (1954, p. 69), are: (1) total concentration of soluble salts, (2) relative proportion of sodium to other cations, (3) concentration of boron or other elements that may be toxic, and (4) the excess of equivalents of bicarbonate over equivalents of calcium plus magnesium.

High concentrations of dissolved salts in irrigation water may cause a buildup of salts in the soil solution, and may make the soil saline. The increased salinity of the soil may drastically reduce crop yields by decreasing the ability of the plants to take up water and essential plant nutrients from the soil solution. The tendency of irrigation water to cause a high buildup of salts in the soil is called the salinity hazard of the water. The specific conductance of the water is used as an index of the salinity hazard.

High concentrations of sodium relative to the concentrations of calcium and magnesium in irrigation water can adversely affect soil structure. Cations in the soil solution become fixed on the surface of the soil particles; calcium and magnesium tend to flocculate the particles, whereas sodium tends to deflocculate them. This adverse effect on soil structure caused by high sodium concentrations in an irrigation water is called the sodium hazard of the water. An index used for predicting the sodium hazard is the sodium-adsorption ratio (SAR), which is defined by the equation:

$$SAR = \frac{Na^+}{\sqrt{\frac{Ca^{++} + Mg^{++}}{2}}},$$

where the concentrations are expressed in equivalents per million.

The U.S. Salinity Laboratory Staff has prepared a classification for irrigation waters in terms of salinity and sodium hazards. Empirical equations were used in developing a diagram, reproduced in modified form as Figure 9, which uses SAR and specific conductance in classifying irrigation waters. This classification, although embodying both research and field observations, should be used only for general guidance because many additional factors (such as availability of water for leaching, ratio of applied water to precipitation, and crops grown) also affect the suitability of water for irrigation. With respect to salinity and sodium hazards, waters are divided into four classes-low, medium, high, and very high. The classification range encompasses those waters which can be used for irrigation of most crops on most soils as well as those waters which are usually unsuitable for irrigation.

Representative data from analyses of water from Twin Buttes Reservoir and the percentage of time that the specific conductance exceeded the indicated



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value for the Colorado River near San Saba and at Wharton are shown in Figure 9. The data show that the sodium hazard for water of the Colorado River basin is low and that the salinity hazard generally is medium.

In the lower Colorado River basin, great quantities of surface water are used for irrigation of rice and grain sorghums. Surface water of the lower basin is excellent for irrigation of these crops.

Geographic Variations in Water Quality

Variations of dissolved solids, hardness, and chloride in the streams in the Colorado River basin are shown in Figures 11, 12, and 13. These values are based on the discharge-weighted average concentrations, as calculated from chemical-quality data. The discharge-weighted average represents approximately the chemical character of the water if all the water passing a point in the stream were impounded in a reservoir, and mixed, with no adjustments for evaporation, rainfall, or chemical changes that might occur during storage. For many of the streams, chemical-quality data are limited, especially data on the chemical quality of flood flows. All the streams will at times have concentrations exceeding those shown, but the averages shown on the maps are indicative of the type of water that would be stored in a reservoir.

Dissolved Solids

The concentration of dissolved solids in surface water of the Colorado River basin is shown on Figure 11. Water of Lake J. B. Thomas in the upper part of the basin contains slightly more than 250 ppm dissolved solids. Below Lake J. B. Thomas is an area of saline inflow that badly degrades the water of much of the upper basin. Part of the inflow is definitely the result of oil-field operations. Most of the remainder of the basin yields surface water averaging less than 250 ppm in dissolved-solids content, but the effect of the saline inflow keeps the average concentration in the main stem above 250 ppm throughout its length (Figure 11). Higher concentrations are found in Beals Creek and Concho River subbasins because of oil-field operations.

The discharge-weighted average concentrations of dissolved solids of the Colorado River near San Saba and at Wharton for the period 1948-65 were 295 and 255 ppm, respectively. The analyses showing annual maximum and minimum dissolved-solids concentrations and the weighted averages for the stations are shown in Table 5.

Time-weighted averages are usually higher than discharge-weighted averages. The duration curves (Figure 7) for concentration of dissolved solids for the Colorado River near San Saba and at Wharton show that 400 ppm dissolved solids has been equaled or exceeded 50 percent of the time at San Saba and that 280 ppm was equaled or exceeded 50 percent of the time at Wharton.

Hardness

Surface water of the Colorado River basin generally is moderately hard (61 to 120 ppm) to very hard (over 180 ppm). Water in Lake J. B. Thomas is moderately hard. From Lake J. B. Thomas to the mouth of Pecan Bayou, the

Colorado River and its major tributaries contain very hard water (Figure 12). Pecan Bayou and most of the downstream tributaries contribute water that is hard (121 to 180 ppm), and the Colorado River contains hard water from the mouth of Pecan Bayou to the coast.

Chloride

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The concentration of chloride in surface waters of the Colorado River basin ranges from less than 50 ppm to over 500 ppm (Figure 13). Water of Lake J. B. Thomas contains less than 50 ppm chloride. Below Lake J. B. Thomas, where oilfield brine and some natural saline flow is reaching the streams, chloride concentrations of several thousand ppm are common. Most of the remainder of the basin yields surface water averaging less than 50 ppm chloride, but the effects of the saline inflow in the upper basin keeps the average concentrations are found in the South Concho River and in the upstream portion of Pecan Bayou probably because of oil-field operations.

Other Constituents

Other constituents of importance in the evaluation of the quality of a water include silica, sodium, bicarbonate, sulfate, fluoride, and nitrate.

Most of the streams in the Colorado River basin contain less than 15 ppm silica, and the annual weighted-average concentration of the Colorado River has usually been less than 10 ppm.

Sodium concentrations are generally less than 50 ppm in most of the streams. In those waters having high chloride concentrations, sodium occurs in quantities approximately equivalent to the chloride. It is therefore present in highest concentrations in the Colorado River in the Ira-Colorado City area. The annual weighted-average concentration of the Colorado River at Austin and Wharton is usually less than 50 ppm.

Bicarbonate is the principal anion in water draining rocks of Cretaceous age. The water of the Concho, San Saba, Llano, and Pedernales Rivers has bicarbonate as the principal anion; concentrations generally range between 200 and 300 ppm. In the lower part of the basin, water draining the younger formations contains much smaller concentrations. The weighted-average concentrations of bicarbonate for the 18-year period 1948-65 for the daily sampling stations on the Colorado River near San Saba and at Wharton are 156 ppm and 164 ppm, respectively.

Sulfate concentrations are generally less than 50 ppm in most of the streams in the basin, although higher concentrations are found in the polluted streams. Although concentrations of over 3,000 ppm are not uncommon, the weighted-average concentration for the Colorado River at Colorado City has ranged from 42 to 456 ppm. The weighted-average concentration for the Colorado River has ranged from 16 to 70 ppm near San Saba and 18 to 45 ppm at Wharton. Fluoride concentrations seldom exceed 1.0 ppm and generally range from 0.2 to 0.4 ppm. Nitrate concentrations are generally less than 3.0 ppm in most of the streams in the basin.

Water Quality in Reservoirs

The principal reservoirs in the Colorado River basin were sampled during the reconnaissance and the chemical analyses are given in Table 6. Analyses are also available for some of the small reservoirs used for public supply (Sundstrom and others, 1949).

Lake J. B. Thomas.--Lake J. B. Thomas, in the upper Colorado River basin, contains water of good quality. Ten analyses during the period 1953-65 show that the water usually contains about 250 ppm dissolved solids, about 25 ppm chloride, and about 60 ppm sulfate.

Lake Colorado City and Champion Creek Reservoir.--Owned and operated by the Texas Electric Service Company, these reservoirs provide cooling water for a steam-electric plant and the municipal supply for Colorado City. The chemical quality of the water in the two reservoirs is similar, about 300 ppm dissolved solids and about 40 ppm chloride.

<u>Robert Lee Reservoir</u>.--Construction of the dam that will form Robert Lee Reservoir began late in 1966. The reservoir site is on the Colorado River downstream from an area of saline inflow in Mitchell and Scurry Counties. The Colorado River Municipal Water District is building the reservoir, and plans to catch the highly mineralized low flow of the river upstream from the reservoir for use in waterflooding projects in several oil fields in the area. This saltwater alleviation program will greatly improve the quality of the water stored in the reservoir, but the chloride content of the stored water probably will, at times, exceed the limits recommended by the U.S. Public Health Service.

Oak Creek Reservoir.--The quality of the water in Oak Creek Reservoir, though still good, has deteriorated slightly since 1953. An analysis in 1953 showed 1.5 ppm chloride and 127 ppm dissolved solids. The most recent analysis in 1965 showed 29 ppm chloride and 239 ppm dissolved solids.

<u>Twin Buttes Reservoir</u>.--Twin Buttes Reservoir was completed in 1962 but because of drought conditions had not impounded much water during this study. For the period that analyses are available (September 1964 to August 1965) the dissolved-solids content of the water has ranged from about 400 to 700 ppm. The quality of the water in Twin Buttes Reservoir is probably adversely affected by oil-field operations in the South Concho River and Spring Creek drainage areas, but the concentrations measured during this study are higher than can be expected when the reservoir is filled.

Lake Nasworthy.--Lake Nasworthy is just downstream from Twin Buttes Reservoir and most of its inflow is water released or pumped from Twin Buttes. Therefore, the water in Lake Nasworthy is similar in chemical quality to that stored in Twin Buttes Reservoir. Two analyses in 1965 showed 451 and 500 ppm dissolved solids.

San Angelo Reservoir.--San Angelo Reservoir, on the North Concho River, impounds water of very good quality; dissolved-solids content has usually been less than 200 ppm.

<u>Coleman Reservoir</u>.--Coleman Reservoir was not impounding water during this study but the probable quality can be inferred from chemical analyses of Jim Ned Creek (site 71, Table 6). Dissolved-solids content of Jim Ned Creek near Coleman has ranged from 130 to 433 ppm and averaged about 200 ppm. Hords Creek Reservoir.--Hords Creek Reservoir contains water of excellent quality, averaging about 15 ppm chloride and 150 ppm dissolved solids.

<u>Brownwood Reservoir</u>.--The water in Brownwood Reservoir is always of good quality as shown by analyses of samples from the Brown County Water Improvement District No. 1 Canal (site 78, Table 6). The dissolved-solids concentration of water drawn from the lake has ranged from 166 to 241 ppm.

Brady Creek Reservoir.--Brady Creek Reservoir was built to provide a municipal water supply for the city of Brady. The water is of excellent quality, usually containing less than 200 ppm dissolved solids.

Lake Buchanan, Inks Lake, Lake Lyndon B. Johnson, Marble Falls Lake, Lake Travis, and Lake Austin.--As a result of the successive impoundment, the quality of the water in these reservoirs is very similar. The Texas State Department of Health has sampled the outflow from Lake Buchanan and Lake Travis since 1957, and the Geological Survey has sampled the outflow from Lake Austin on a daily basis since October 1947. The analyses show that the water is always of good quality. Calcium and bicarbonate are the predominant ions, and dissolved-solids content is usually between 250 and 350 ppm.

Decker Lake.--Construction of Decker Lake began in 1966 and no water was impounded during this study. The reservoir, when completed, will store water pumped from the Colorado River to be used for cooling at the city of Austin's Decker Creek steam-generating plant. The water will be diverted from the river downstream from the Austin sewage outfall. The dissolved-solids content of the water should range from 250 to 350 ppm, but the organic quality of the water may at times be poor.

Lake Bastrop.--Lake Bastrop is a Colorado River off-channel reservoir that stores cooling water for a Lower Colorado River Authority steam-generating plant. Chemical analyses are not available, but the water is similar to that of the Colorado River passing Austin, which usually contains from 250 to 350 ppm dissolved solids.

Eagle Lake.--Eagle Lake is a Colorado River off-channel reservoir owned by the Lakeside Irrigation Company. During flood flows, water is pumped from the Colorado River and stored until needed for irrigation. An analysis in 1959 showed the water to be of good quality; the dissolved-solids concentration was 176 ppm.

Water Quality at Potential Reservoir Sites

One of the purposes of the reconnaissance was to appraise the quality of the water which will be available for storage at potential reservoir sites. Many sites studied by various Federal, State, and local agencies are indicated on Figure 4.

<u>Stacy</u>.--A reservoir on the Colorado River at the Stacy site would impound water from the Colorado and Concho Rivers. The water of the Colorado River impounded at the Stacy site would contain slightly more than 500 ppm dissolved solids.

Upper Pecan Bayou.--A reservoir at the Upper Pecan Bayou site would impound water of good quality; the water would be hard and contain less than 250 ppm dissolved solids.

<u>Brownwood Reservoir (Enlargement)</u>.--The enlargement of Brownwood Reservoir should not cause any change in the quality of the water stored. The water should still be hard and contain less than 250 ppm dissolved solids.

<u>San Saba</u>.--The quality of the water at the San Saba site can be determined from the chemical quality data for the San Saba River at San Saba. The water that would be stored will be very hard but contain less than 250 ppm dissolved solids.

<u>Mason</u>.--The quality of the water that could be impounded at the Mason site on the Llano River can be inferred from analyses of the Llano River at Junction and Llano. The water available for storage is calcium bicarbonate in type and is hard; the dissolved-solids concentration is usually less than 250 ppm.

<u>Pedernales</u>.--According to periodic chemical-quality data for the Pedernales River near Johnson City, water impounded at the Pedernales site would be hard and contain about 250 ppm dissolved solids.

<u>Columbus Bend</u>.--Water available at this site would be very similar in quality to the water sampled at the daily quality station at Wharton where the weighted-average dissolved-solids concentration has ranged from 198 to 328 ppm.

<u>Matagorda</u>.--Water available at the Matagorda site would also be very similar to the water sampled at the daily quality station at Wharton.

Problems Needing Additional Investigation

This reconnaissance of the chemical quality of the Colorado River basin has shown that the natural runoff of the basin is generally of good chemical quality.

However, saline inflow principally from oil-field operations makes the water in the upper part of the basin unfit for most uses and increases the salt load in the main stem throughout its length. Continuing study will be necessary to evaluate the salt-water alleviation program that is planned for the area above the Robert Lee Reservoir site. Small areas in the Concho River and Pecan Bayou subbasins are slightly polluted with salt water produced with oil or gas.

A potential water-quality hazard exists in the portion of the Colorado River basin that is usually considered noncontributing. In the drainage area of Beals Creek just upstream from Big Spring, natural and oil-field brines are impounded in a depression called Natural Dam Salt Lake (site 19, Table 6). The impounded water may contain more than a hundred thousand ppm of dissolved solids, and the lake bed is covered by a thick layer of deposited salts. Levees have been built as a precaution against overflow. If the lake should overflow the small amount of brine in storage would not cause serious pollution in downstream reservoirs, but many tons of the deposited salt in the lake would be dissolved and carried downstream, thereby greatly increasing the possibility of serious water-quality damage. Organic quality is generally good throughout the basin; however, some concern is being expressed regarding bacterial contamination of the Highland Lakes by septic-tank effluent in areas of housing developments, and by waste discharges from pleasure boats.

Continued municipal and industrial growth in the basin will cause an increase in the waste-disposal burdens of the stream system. Meanwhile, the impoundment of water in upstream reservoirs will cause a reduction of streamflow now utilized for the assimilation of municipal wastes. Consequently, continued municipal and industrial growth will require that wastes be consistently treated to the maximum extent if gross pollution of streams is to be avoided in the future.

Impoundment of water will likewise result in some changes of water quality. Beneficial effects will include: the reduction in turbidity, silica, color, and coliform bacteria; the evening-out of sharp variations in chemical quality; the entrapment of sediment; and a reduction in temperature. On the other hand, detrimental effects of impoundment will include: an increase in the growth of algae; the reduction of dissolved oxygen; and an increase of dissolved solids as a result of evaporation. The continued extensive development of the water resources of the Colorado River basin will necessitate detailed study of the changes in water quality.


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Water year	U.S.G.S. Water-Supply Paper No.	T.W.D.B. Report No.	Water year	U.S.G.S. Water-Supply Paper No.	T.W.D.B. Report No.
1940-45		*1938-45	1955	1402	*1955
1946	1050	*1946	1956	1452	Bull. 5905
1947	1102	*1947	1957	1522	Bull. 5915
1948	1133	*1948	1958	1573	Bull. 6104
1949	1163	*1949	1959	1644	Bull. 6205
1950	1188	*1950	1960	1744	Bull. 6215
1951	1199	*1951	1961	1884	Bull. 6304
1952	1252	*1952	1962	1944	Bull. 6501
1953	1292	*1953	1963	1951	Rept. 7
1954	1352	*1954			

* "Chemical Composition of Texas Surface Waters" was designated only by water year from 1938 through 1955.

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Year	Water-Supply Paper No.	Year	Water-Supply Paper No.	Year	Water-Supply Paper No.
1898	28	1919	508	1940	898
1899	37	1920	508	1941	928
1900	50	1921	528	1942	958
1901	75	1922	548	1943	978
1902	84	1923	568	1944	1008
1903	99	1924	588	1945	1038
1904	132	1925	608	1946	1058
1905	174	1926	628	1947	1088
1906	210	1927	648	1948	1118
1907	248	1928	668	1949	1148
1908	248	1929	688	1950	1178
1909	268	1930	703	1951	1212
1910	288	1931	718	1952	1242
1911	308	1932	733	1953	1282
1912	328	1933	748	1954	1342
1913	358	1934	763	1955	1392
1914	388	1935	788	1956	1442
1915	408	1936	808	1957	1512
1916	438	1937	828	1958	1562
1917	458	1938	858	1959	1632
1918	478	1939	878	1960	1712

The following U.S. Geological Survey Water-Supply Papers contain results of stream measurements in the Colorado River basin, 1898-1960:

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Table 4.---Index of surface-water records in the Colorado River basin

Refer-		Drainage			Calen	dar Years			
ence no.	Stream and Location	Area (sq. miles)	01-1061	1911-20	1921-30	1931-40	1941-50	1951-60	1961-70
г	Lake J. B. Thomas near Vincent	934							2
7	Bull Creek near Ira	388							
n	Bluff Creek near Ira	42.6							
4	Colorado River near Ira	1027							
2 2	Deep Creek near Snyder								
9	Deep Creek near Dunn	188							
7	Sulphur Creek near Dunn							3	
80	Colorado River near Cuthbert	1428							81
6	Colorado River at Colorado City	1482			NANNA NANNNA NANNNA NANNNA NANNNA NANNNA NANNNA NANNNA NANNNA NANNNA NANNNA NANNNA NANNNA NANNNA NANNNA NANNNA NANNNA NANNNA NANNNA NANNNA NANNNANNNA NANNNA NANNNANNNA NANNNA NANNNANNNA NANNNA NANNNANNNA NANNNANNNA NANNNANNNA NANNNANNNA NANNNANNNNA NANNNNNN				
10	Morgan Creek near Westbrook	228							
11	Graze Creek near Westbrook	21.2							
12	Morgan Creek near Colorado City	262							
13	Lake Colorado City near Colorado City	267							2
14	Champion Creek near Colorado City	158							
15	Champion Creek Reservoir near Colorado City	203							1
16	Sulphur Springs Creek near Big Spring								2.
17	Calf Creek near Stanton								8
18	Buzzard Creek at U.S. Highway 87 near Big Spring								2
19	Natural Dam Salt Lake near Big Spring								ž
20	Beals Creek above Big Spring	494							23 23 23 23 23 23 23 24 24 24 24 24 24 24 24 24 24 24 24 24
Disch Peri	arge <u>community</u> Gage heights only <u>unitarianterinant</u> odic discharge measurements <u>community</u> Daily	Gage he y chemical qu	ights and discharg ality	je medsuremei Perio	nts <u>states</u> dic chemical quality	Reservoir cont-	ents <u>Earne</u> Water temperature		

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urements ANALY Reservoir contents	ge heights and discharge measu	900	harge Announce Gage heights only Announcement	Disch
		396	Spring Creek above Tankersley	40
			Spring Creek Springs near Mertzon	39
		1128	Middle Concho River near Tankersley	38
		1381	Middle Concho River above Tankersley	37
		344	South Concho River at Christoval	36
			South Concho Irrigation Co.'s canal at Christoval	35
			Anson Springs near Christoval	34
			Dry Creek near Christoval	33
		471	Elm Creek at Ballinger	32
			Lake Winters near Winters	31
		5240	Colorado River at Ballinger	30
			Oak Creek near Blackwell	29
		222	Oak Creek Reservoir near Blackwell	28
			Colorado River near Bronte	27
			Mountain Creek Reservoir at Robert Lee	26
		4170	Colorado River at Robert Lee	25
		3880	Colorado River near Silver	24
		973	Beals Creek near Westbrook	23
			Coahoma Draw Tributary near Big Spring	22
		515	Beals Creek at Big Spring	21
Calendar Years 1921-30 1931-40 1941-50 1951-60 1961-70	01-1061	Drainage Area (sq. miles)	Stream and Location	Refer- ence no.

Periodic discharge measurements Daily chemical quality Periodic chemical quality Water temperature

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Refer	Ctream and Location	Drainage			Calendar	Years				
ençe Do	Stream and Location	Area (sq. miles)	01-1061	1911-20	1921-30	1931-40	1941-50	1951-60	1961-70	
41	Dove Creek Springs near Knickerbocker									
42	Dove Creek near Knickerbocker								í 	
43	Dove Creek at Knickerbocker	198						28		
44	Spring Creek near Tankersley	734								
45	Twin Buttes Reservoir near San Angelo	2546								_
46	Pecan Creek near San Angelo	84.9								
47	Tom Green Co. WCID No. 1 Canal near San Angelo								1	
48	Lake Nasworthy near San Angelo	2507							ł	
49	South Concho River at San Angelo	2535								
50	Quarry Creek near Sterling City						1			
51	North Concho River at Sterling City	539								
52	Broome Creek near Broome									-
53	North Concho River near Carlsbad	1144								
54	Nolke Station Creek near San Angelo									
55	Gravel Pit Creek near San Angelo									
56	San Angelo Reservoir at San Angelo	1383							2	
57	North Concho River at San Angelo	1402								
58	Concho River near San Angelo	4097								-
59	Concho River near Paint Rock	5132								
60	Mukewater Creek Subwatershed No. 9 near Trickham	4.02								
Disch	rrge communications Gage heights only municipation	Gage	heights and disch	arge measurements		Reservoir conten	s		-	

Periodic chemical quality management Water temperature

Daily chemical quality

Periodic discharge measurements

Table 4.---Index of surface-water records in the Colorado River basin---Continued

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Table 4.---Index of surface-water records in the Colorado River basin---Continued

Refer		Drainage			Calendar	Years			
ence no.	Stream and Location	Area (sq. miles)	01-1061	911-20	1921-30	1931-40	1941-50	1951-60 1961-70	
61	Mukewater Creek at Trickham	70.0							
62	Colorado River at Winchell (Milburn)	12680				33927			
63	Deep Creek Subwatershed No. 1 near Placid							2	
64	Deep Creek Subwatershed No. 2 near Placid							2	
65	Deep Creek Subwatershed No. 3 near Placid	3.42							
99	Deep Creek Subwatershed No. 4 near Placid							<u> </u>	
67	Deep Creek Subwatershed No. 5 near Placid								
68	i Deep Creek near Mercury	43.9							
69	Deep Creek Subwatershed No. 8 (Dry Prong Deep Creek) near Mercury	4.32							
70	Dry Prong Deep Creek near Mercury	8.31							
71	Jim Ned Creek near Coleman	333	· · · · · · · · · · · · · · · · · · ·						
72	Hords Creek Reservoir near Valera	48							
73	Hords Creek near Valera	53							
74	i Hords Creek at Coleman	107				2	×		
76) North Fork Pecan Bayou near Clyde								
76) Pecan Bayou at Burkett							S #	
17	Pecan Bayou at FM Road 2559 near Brownwood								
78	i Brown Co. WID No. 1 Canal near Brownwood								
52	Brownwood Reservoir near Brownwood	1535							
8() Pecan Bayou at Brownwood	1614		3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3					
Disc Per	harge community Gage heights only community Dai	Gage I ily chemical q	heights and discharge m wality means	easurements Periodic c	hemical quality S	Reservoir con	tents <u>Econom</u> Water temperat	ure (2000000000000000000000000000000000000	

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Table 4.--Index of surface-water records in the Colorado River basin--Continued

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Refer		Drainage			Calend	ar Years			
ence no.	Stream and Location	Area (sq. miles)	01-1061	1911-20	1921-30	1931-40	1941-50	1951-60	1961-70
81	Pecan Bayou near Goldthwaite								
82	Springs at Fort McKavett								
83	Noyes Canal at Menard								
84	San Saba River at Menard	1151							
85	Brady Creek near Eden	97							
86	i Hardin Creek at Eden							P.	
87	Brady Creek Reservoir near Brady	513							
88	Brady Creek at Brady	575							
89	San Saba River at San Saba	3042							
06) San Saba Springs at San Saba								
16	Colorado River near San Saba	18700							
92	Buchanan Reservoir near Burnet	19350							2
93	North Llano River near Junction	914							
94	South Llano River near Telegraph								
95	Seven Hundred Springs near Telegraph								2
96	Llano River near Junction	1874							
26	Beaver Creek near Mason	218							
98	Llano River near Castell	3747							
66	Llano River at Llano	4233							
100	Colorado River at Marble Falls								
Disch	harge annummung Gage heights only annummun	Gage	e heights and disch	arge measurement		Reservoir conter	its		
Per	iodic discharge measurements	ly chemical	quality	Periodi	c chemical quality		Vater temperature	000000000000000000000000000000000000000	

Table 4.--Index of surface-water records in the Colorado River basin--Continued

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Refer-		Drainage			Calenc	tar Years			
ence no.	Stream and Location	Area (sq. miles)	01-1061	1911-20	1921-30	1931-40	1941-50	1951-60	1961-70
101	Pedernales River at Stonewall	647							
102	Salt Branch at Stonewall								8
103	Pedernales River near Johnson City	947							
104	Pedernales River near Spicewood	1294							
105	Lake Travis near Austin	26230							
106	Bull Creek at Doernige Park near Austin							2	
107	Lake Austin at Austin								Т
108	Colorado River at Dam above Austin								
109	Barton Creek at Hays Co. line near Dripping Springs								
110	Barton Creek above Barton Springs at Austin						X		
111	Barton Springs at Austin								
112	Waller Creek at 38th Street at Austin	2.31							
113	Waller Creek at 23rd Street at Austin	4.13							
114	Colorado River at Austin	26500 N							
115	Little Walnut Creek near Austin								
116	Onion Creek near Driftwood								
117	Onion Creek at Buda								
118	Fox Branch near Oak Hill								
119	Onion Creek near Del Vallc	337							
120	Onion Creek below Del Valle								
Discl	iorge <u>communitations</u> Gage heights only <u>animummunitation</u>	Gagi	e heights and discharge I	neasurements		Reservoir conte	ents		
Per	iodic discharge measurements	y chemical	l quality	Periodic	chemical quality		Water temperature	6	_

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Table 4.--Index of surface-water records in the Colorado River basin--Continued

Refer		Drainage				Caler	ndar Years				
ence no	Stream and Location	Area (sq. miles)	-1061	0	911-20	1921-30	1931-40	1941-50	1951-60	1961–70	
121	Wilbarger Creek near Pflugerville	4.61									
122	Colorado River at Bastrop	27 500				8 1101111111111111111111111111111111111					
123	Cedar Creek near Bastrop										
124	Piney Creek near Bastrop										
125	Colorado River at Smithville	27980						XX			
126	Dry Creek at Buescher Lake near Smithville	1.48									
127	Colorado River at LaGrange	28530									
128	Redgate Creek near Columbus	173									
129	Cummins Creek near Columbus								1		
130	Colorado River at Columbus	29170									
131	Colorado River near Eagle Lake	29270						3	2		
132	Eagle Lake at Eagle Lake								1		
133	Colorado River at Wharton	29480									
134	Colorado River near Bay City	29750					3				
Disch	torge terrorisments Gage heights only munumment	Gage	e heights a	nd discharge n	neasurements		Reservoir cont	ents			
Per	iodic discharge measurements treet	ily chemical	quality I		Periodic	chemical quality		Water temperatu	re <u>kooooooooooooooo</u>		

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(Analyses listed as maximum and minimum were classified on the basis of the values for dissolved solids only; values of other constituents may not be extremes.) Results in parts per million except as indicated

				. ~	m .					10.10 m m /c					
	H		7.6	- 2.5	37	· ·		8.2	8.1		_	8.2	8.3	01014 01014	5.77 5.4.4
Specific conduct-	ance (micro- mhos at 25° C)		5,510	427 3,500	413	4,450	236	2,100	354	11,500 4,390 10,300 5,460 1,010		3,960	310	2,900 3,720 632 1,880 1,880	5,710 4,100 506
So-	adsorp- tion ratio		11			1	: :	6.2	1.7	37 17 30 16 5.3		1	ł	1,3250 1,3250 1,425 1,42	6.8 5.4 1.8
Per-	cent so- dium		39 39 39	49 67	47 51	5	12	61	49	87 83 75 69		73	10 00	41 47 36 36	443 433 433
CO,	Non- carbon- atc		381 0	456 0	09	, I	420	231	0	706 382 448 74		314	2	658 926 360 72	1,560 1,110 41
Hard as Co	Cal- cium, magne- sium		438 79	106 584	110 124			400	86	782 426 948 708 140		526	102	941 1,140 552 150	1,770 1,270 136
ds	Tons per day		3.0	21 .5	16 6.2		48	.49	.57						
solved soli	Tens per acre- foot		4.22	. 30 2.92	.34	1	3.01	1.66	-29	9.51 3.39 8.69 4.46 .73		3.18	.28	2.52 3.16 .46 1.48 .37	5.00 3.55 .39
Dis	Parts per mil- lion		3,100	206 2,150	248 302	I	2,210	1,220	210	$ \begin{array}{c} 6,970\\ 6,390\\ 3,280\\ 3,280\\ 537 \end{array} $		2,340	209	1,850 2,320 337 1,090 272	$^{3,680}_{2,610}$
Bo-	(B)														
Ni-	trate (NO.)		0.00	3.6 1.0	2.5 1.6	ł	3.2	3.0 3	4.0	16 4.0	A	1.5	5.4	$1.3 \\ 1.3 \\ 1.8 \\ 1.0 $	1.2
Fluo-	ride (F)	AR IRA									CAR IR	I	0.3	440	.
Chlo-	ride (CI)	REEK NEA	1,580 11	40 690	38 09	980	11 780	378	36	3,790 1,450 3,520 1,620	CREEK NI	930	22	570 860 110 375 78	1,680 1,050 70
Sul-	fate (SO.)	BULL C	317 17	57 649	32 46	1		304	24	489 334 282 67	BLUFF	423	26	478 534 21 211 42	606 564 43
Bicar-	bonate (HCO ₃)	2.	69 114	156	142 144	ł		207	108	92 54 317 81	з. Э.	258	116	346 264 233 96	251 202 116
Po-	tas- sium (K)		23 23 17	1 / 2 549	44 60	1	1862	286	37	370 792 100 145		562	26	297 388 65 39	562 143 48
ŝ	dium (Na)		1.(ณ์ณ์					
Mag-	ne- sium (Mg)		39 4.6	83 0.4	4.8			38	5.2	62 62 68 68 68 7 5 7		55	4.7	88 109 42 7.4	169 124 6.4
Cal.	(Ca)		111 24 26	70 6	36 36	1	18	86	26	211 225 172 42		120	33	232 56 152 48	430 305 44
	Iron (Fe)														
	Silica (SiO.)		11	0. 0. 0.	16 18	ł	7.0	6.2	11	5.2 16 .5 .6 6.6		17	13	15 7.5 9.4	13 22 19
Mean	dis- charge (cfs)		0.36 110 303	60.	24.0 7.43	.01	13.8	.15	1.01	a.91 a.10 a.72		-	1	a0.06 a.03 8.65 a.1 a.0	a.01 a.02 3.49
	Date of collection		Apr. to Sept., 1950 Maximum, Aug. 16-20 Mionum, May 2	Water year 1951 Maximum, 1951- May 1-10, 1951-	Multur, July 1-7 Weighted average-	Water year 1952 Maximum, August 1-31, 1952	Winnmum, Sept. 22-28	Water year 1953 Maximum, Mar. 1-31, 1953	Dec. 18-31,1952	Oct. 11, 1959 Sept. 17, 1964 Dec. 21, 1964 Jan. 28, 1965 May 11, 1965		Water year 1950 Maximum, Aug. 30, Sept. 1, 1950 Misimum M. 11-	13,	Mar. 24, 1964 Apr. 27, 1964 May 30, 1964 June 2, 1964 Scpt. 17, 1964	Nov. 17, 1964 Jan. 28, 1965 May 17, 1965

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7.1 6.4 7.8 7.0 7.5 7.7 6.8 7.3 7.0 7.5 6.6 7.8 6.9 7.4 7.1 6.9 Ηd 467 $^{411}_{1,100}$ 50,300 465 1,170 1,7205,340 Specific conduct-ance (micro-mhos at 25° C) 49,900 000 040 900 $^{74}_{7,600}$ 80,200 663 3,260 69,100 53,400 ŀ~ 64.0 $^{2.3}_{20}$ So-dium adsorp-tion ratio 113 5.5 26 3.17.6 95 8.7 18 3.5 $\frac{3.1}{10.4}$ 94 --4.9 60 111 Per-cent so-dium 88 53 86 86 86 88 65 77 88 60 78 8811 87 76 --83 59 Non-carbon-ate 4,970 40 370 3,880 15 45 6,40028 489 3,730 13 850 091 98 365 590 26 209 4,640Hardness as CGCO. ŝ . ດ Cal-cium, magne-5,050158 474 5,650 114 303 $102 \\ 592$ 4,750 $196 \\ 464$ 500 139 581 19.0 3,830 5,990 69 127 960
 101
 140ium ê, e, 52.1 37.2 2.87 428 26.2 1,840 21.4 10.9 12.0 111 762 Tons per day Dissolved solids .35 74.9 .81 5.34 55.3 .33 .99 1.30 93.3 .55 6.22 82.3 .45 2.71 32 54.682.1 N Tons per acre-foot 62. 255 4,990 53,100 592 3,930 57,900
 328
 1,99039,100 65,600 403 4,570 57,900 600 246 725 954 400 234 300 Parts per mil-lion 39, 44, ŝ 2.8 1.5 5.5 2.0 4.0 1.0 2.5 Ł ł Ni-trate (NO.) IRA NEAR $\left\{ \cdot \right\}$ $\{ \cdot \} \in \{ \cdot \}$ 1 71 141 Fluo-ride (F) ł 1 11 $\{ \ | \ | \ |$ 111 RIVER $^{69}_{2,640}$ 900 230 060 23,700 $^{428}_{1,740}$ 35,600 139 2,390 31,600 107 1,010 200 20065 260 $^{200}_{74}$ Chlo-ride (Cl) 21, 28, 31, 21, COLORADO 4,08058 310 4,7504,360 37 168 020 27 406 380 23 80 840 95 314 190 34 397 $^{23}_{74}$ Sul-fate (SO.) ŝ ŝ ñ ιŋ, Bicar-bonate (HCO.) 4 172 99 136 130 119 121 69 108 115 129 108 100 98 144 127 92 105 146 119 1136 (K) fas-53 1,670 18,400 159 1,320 19,800 $^{22},600$ 96 $^{1},520$ 500 60 198 600 53 220 15,100 280 20,000 76 638 ł So-dium (Na) 13, Ë. 22, 573 5.8 23 10 664 5.8 49 4.1 50 00 75 Mag-ne-sium (Mg) 353 500 10 ei 30 389 7. 10 583 $^{10}_{37}$ 475 $1,200 \\ 47 \\ 124$ 955 34 155 440 23 944 28 40 ,120 62 125 510 46 152 320 36 84 Cal-cium (Ca) Iron (Fe) 12 9.7 3.4 $^{4.8}_{16}$ 8.2 3.3 8.0 8.2 7.8 11 9.6 Silica (SiO_i) 4.9 1.2 8.3 8.3 1.816 0.18 75.7 2.76 .02 268 2.47 .07 2910 43.5 $^{131}_{41.1}$ 296 4.1 $^{.1}_{4.1}$ 366 8.2 Mean dis-charge (cfs) 7 Mater year 1963 Maximum, Apr.19-23, 25-26,1965-Majnimum, May 22-23, 1963 Weighted average-Water year 1965 Maximum, May 1-6, 1965--Minimum, May 14,16 : Weighted average-Water year 1959 Maximum, Mar. 14-25, 1959-----Minimum, June 4-6, 1959------Weighted average-Water year 1961 Maximum, May 1-6, 1961--Minimum, Oct. 19, 1960--Weighted average-Water year 1962 Maximum, Maril2-25,1962 Minimum, June 12-Weighted average-Water year 1964 Maximum, May 1-2, 1964--Minimum, Aug. 27--Weighted average-Apr. 16-30,1960 Minimum, July 5-6-Weighted average-Date of collection Water year 1960 Maximum,

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	Hq		6.8 8.0 7.7		::::	1		ł		7.5		6.9	7.8	7.7 7.8	6.6	7.7
Specific conduct-	ance (micro- mhos at 25° C)		37,100 358 1,330		21,300 601 1,540	20,700	342 1,230	18,300	274 916	21,500	$^{484}_{1,670}$	26,100	$^{482}_{1,330}$	28,300 568 3,570	42,500	1,140 7,620
So- dium	adsorp- tion ratio		0.9 4.5		111	ł		ł		ł	1 1	ł	11		70	5.4 25
Per-	so. dium		52 52 52		82 57 74	83	46 72	83	36 69	83	68 76	84	65 74	83 56 80	85	69 83
ress CO.	Non- carbon- ate		3,500 7 82		1,910 29 84	1,800	0 73	1,570	32 0	1,890	0.68	2,300	0 09	2,530 239 239	3,740	46 587
Hardr as Co(Cal- cium, magne- sium		3,570 130 215		2,010 118 182	1,890	88 167	1,660	89 128	1,990	71 188	2,380	75 161	2,610 120 346	3,820	672
8	Tons per day				84 65	52,900	3,630 287	29	74 228	9.3	440 163	50	778 173	36 1,410 264	8.]	324 56.0
olved solid	Tons per acre- foot		39.3 .29 1.03		17.5 1.15	18.4	. 98	15.6	.70	18.8	.39	23.3	.36	25.1 .45 2.77	40.8	.78
Diss	Parts per mil- lion		28,300 215 755		12,900 344 847	13,500	206 738	11,500	176 532	13,800	286 948	17,100	264 742	18,500 332 2,040	30,000	571 4,640
ġ	I (B)															
Ni-	trate (NO.)	IBERT	4.2	CITY	3.9	4.0	1.8	3.0	2.2	2	2.28	ł	1.8	5.5	1	6.9
Fluo-	ride (F)	AR CUTH		ORADO												
Chlo-	ride (CI)	IVER NE/	15,700 310 310	R AT COI	6,450 111 360	6,900	30 298	5,940	10 182	7,100	60 408	8,860	60 303	9,570 58 982	15,900	$^{217}_{2,390}$
Sul-	fate (SO,)	ORADO R	2,040 15 70	DO RIVE	1,650 97	1,500	26 85	1,220	30	1,500	39 114	1,790	35 89	2,000 49 222	2,800	62 456
Bicar-	bonate (HCO.)	8. COI	91 150 163	COLOR	11 <i>9</i> 108 120	110	112	108	113	121	116 120	633	116 123	105 167 130	96	125 104
Po-	m tas- sium (K)		310 32 24 198	9.	$\frac{4}{71}, \frac{090}{243}$	4,370	35 201	3,720	23 131	4,450	70 279	5,550	65 214	6,000 71 634	9,930	$152 \\ 1,500$
- S	uip (N		6		10		<u>в</u> .		<u>69</u>		. 4		0,	0,		2
Mag	siun (Mg		11345		1205	1 192	7 12	145	8 0	7 195	69 17 30	2 237	13.5	5 2 6 3 0 2 6 0 2 0 2 6 0 2 0 0 2 0 0 2 0 0 2 0	7 379	0 56
Cal-	cium (Ca)		862 862 862		467	44]	24	425	<u>ର</u> ନ	477	810	26.	2 4	0 0 0 0 0 0	.06	13.5
	(Fe)	-														
	Silica (SiO.)	-	13.3			1				5.8	121		10	2 17.4		7 11
Меап	dis- charge (cfs)		3,250.9 79.6		2.4 70.4 83.4	1,450	6,532 147	e.	156 163	ci.	570 63.6	1.0	1,091	1,576. 1,47.9		210
	Date of collection		Mar. to Sept. 1965 Maximum, Apr. 1-12, 25, 1965- Minimum, May 16 Weighted average-		May to Sept. 1946 Maximum, 1946 May 9, 1946 Minimum, Junc 27- Weighted average-	Water year 1947 Maximum, May 10, 1947	Minimum, may 12-14,17, 1947- Weighted average-	Water year 1948 Maximum, Apr. 23-28, May 1-15, 1948-	Minimum, Oct. 26, 1947 Weighted average-	Water year 1949 Maximum, Apr. 11-18,1949	Minimum, Oct. 11, 1948 Weighted average-	Water year 1950 Maximum, Mar. 1-9,11 -19, Apr.13-15,1350-	Minnum, Sept. 5-9 Weighted average-	Water year 1951 Maximum, Apr.1-30, May 1-7, 1951 Minimum, Aug.23-25 Weighted average-	Water year 1952 Maximum, Mar. 11, 1952	Minimum, July 16-17 Weighted average-

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8.0 --2 7.0 7.2 8.0 7.5 7.6 7.4 7.8 7.1 7.77.6 7.3 7.1 Hq 24,100 270 1,140 38,300 728 4,190 552 1,760 346 946 200 Specific conduct-ance (micro-mhos at 25° C) 800 776 090 800 528 670 800 600 972 190 300 687 300 5 с. С 38. 32 ŝ. 4 34 1 28 ള് 27 5.0 70 4.8 18 9.6 50 --7.1 69 4.1 18 1.6 5.8 4.1 16 ഗഗ So-dium adsorp-tion ratio 69 61.00 3 54 64 59 Per-cent so-dium 84 86 82 82 18 18 86 68 83 811 86 47 72 84 67 80 85 85 82 82 85 54 75 3,220 19 278 2,910 72 49 216 16 200 6 54 Non-carbon-ate 14 290 1,620 490 420 38 296 2,750 26 226 Hardness as CaCO. ro. 0, 2 0 2,290 81 153 490 122 382 ,830 560 114 260
 120
 368
 368020 $112 \\ 202$ 390 1,730 $146 \\ 313$ 310 88 130 Cal-cium, magne-sium ev. esî. ŝ e, e 2 86.7 ,461 .6] 192 .080 .81.9 49.9 15.3 828 100 ł ∞ 797 196 115 040 195 468 110 $\frac{080}{244}$ 68. 144 Tons per day 152solids ÷ N -23.2 .20 ..39 .74 .52 39.5 .62 3.50 .41 90.66 75 30.3 <u>т</u> 17.0 2 C) Dissolved Tons per acre-foot 36. 33. 26. 26. 39. 16,900 150 662 $^{302}_{1,010}$ 000 385 010 500 453 570 000 $\frac{900}{443}$ 400 289 954 800 208 500 542 880 Parts per mil-lion 26, 19. Ξ, ē, 2 28. ં 22 24 Bo-Bo-Bo-COLORADO CITY--Continued 1 0.1 2.2 2.2 1 0.1 Ni-trate (NO) 2.8 1.8.1 111 ł. Fluo-ride (F) 8,900 27 270 15,300 178 1,320 $500 \\ 147 \\ 260 \\ 200$ 46 235 212 922 144 010 800 90 153 100 83 130 800 620 10,100 Chla-ride (Cl) 4. ģ ŝ Ë. ____ 13, 470 45 228 650 18 74 34 30 16 680 1,920 290 33 213 070 500 100 TT $^{28}_{90}$ 50 L73 Sul-fate (SO,) 2 Ļ, 4 e, RIVER Ń 0 ÷ Bicar-bonate (HCO_) 58 111 87 104 104 77 102 105 100 120 135 117 92 125 127 119 124 122 99 102 119 119 COLORADO Po-tas-(K) 66 305 5,550 --187 $140 \\ 583$ 310 $\begin{array}{c} 020\\ 104\\ 795 \end{array}$ 8,150 $62 \\ 277$ 67.0 $35 \\ 151$ 080 330 99 641 530 121 828 6 g ത്. Ŀ. 6 So-dium (Na) ெ 5.0 24 7.7 318 5.2 31 5.8 348 5.5 29 514 246 --42 11 3.3ιO. Mag-ne-sium (Mg) 14.5 290 260 235 169 288 735 39 100 $36 \\ 54$ 36 30 414 46 86 598 35 374 40 102 656 735 569 Cal-cium (Ca) Iron (Fe) 5.0 13 10 4.8 9.6 8.5 6.7 10 9.3 5.4 9.8 3.9 4.5 3.8 12 10 4.4 Silica (SiO.) 113 122 2.85 2.31 2.5 ,700 11.3 20 .84 2.04,607 ,85.5 0.21 692 14.9 .01 78.8 23.6 977 71.9 338 75.8 2 Mean dis-charge (cfs) ,919163 £50. 20. 3 ч Water year 1957 Maximum Jan 24-31, Feb. 1-6, 1957 Minimum, May 12-14, 17-18-------4, Weighted average-Water year 1958 Naximum, Feb. 1-10, 1958 Minimum, May 28-29, June 3-4------Weighted average-Water year 1360 Masimum, June 9-13, 1360 Minimum, July 7--Weikhted average-Water year 1961 Maximum, 1-15, 1961 Minimum, June 15-17,----Weighted average-Maximum, Apr. 5-26, 1962 Minimum,Sept.5-7- 3 Weighted average-Mar. 8-19, 1954 Minimum, May 12-13, 15, 19, 1954---1 Weighted average-Apr. 1-7, 1953-Minimum, July 2-3, 13-14, 1959----Weighted average-Water year 1953 Maximum, Apr.1-14, 1953-Minimum,Aug.19-22 Weighted average-Date of collection Water year 1954 Maximum, Water year 1959 Maximum Water year 1962

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	Mean		1	Cal-	Mag-	So-	Po-	Bicar-	Sul-	Chlo-	Fluo-	Ni-	Bo-	Dis	solved so	lids	Hard as Co	Iness 1CO	Per-	So-	Specific conduct-	
Date of collection	dis- charge (cfs)	(SiO ₂)	(Fe)	cium (Ca)	ne- sium (Mg)	dium (Na)	tas- sium (K)	bonate (HCO ₂)	fate (SO,)	ride (Cl)	ride (F)	trate (NO-)	ren (B)	Parts per mil- lion	Tons per acre- foot	Tons per day	Cal- cium, magne- sium	Non- carbon- atc	cent so- dium	adsorp- tion ratio	ance (micro- mhos at 25° C)	pH
						9.	COLOR	RADO RIV	ER AT C	OLORADO	CITY	-Contin	ued					L				
Water year 1963 Maximum, Apr. 1-26,1963- Minimum,May 23 Weighted average-	2.3 1,650 16.8	3.1 13 9.1		661 34 146	291 11 51	6,: 1,:	340 89 120	118 122 126	1,250 43 368	10,800 123 1,790	0.5	6.1		19,400 380 3,550	26.8 .52 4.83	120 1,690 	2,850 131 572	2,750 31 469	83 60	$52 \\ 3.4 \\ 18$	28,000 714 5,670	7.1 7.6 7.2
Water year 1964 Maximum, Apr. 1-18, 22-25, 1964 Minimum.	. 8	7.0		680	307	7,450	24	104	2,130	11,900				22,500	31.0		2,960	2,870	84		31,100	6.5
Oct. 24, 1963 Weighted average-	182 9.1	$11 \\ 8.1$		40 140	9.7 48	1,	108 230	120 107	$67 \\ 349$	$144 \\ 1,960$.2	5.0		444 3,790	,60 5.15		$140 \\ 547$	42 459	63 81	$\begin{array}{c} 4.0\\ 20.4 \end{array}$	805 6,600	7.6 7.0
Water year 1965 Maximum, Apr. 1-21, 1965 Minimum, May 16 Weighted average-	.7 2,150 41.3	2.4 12 9.3		920 50 89	412 4.4 22	10,600	31 41 179	$97 \\ 143 \\ 148$	2,700 41 153	17,400 49 758				$32,100 \\ 270 \\ 1,580$	44.6 .37 2.15	 	3,990 143 310	3,910 26 189	85 38 63	1.5 8.6	$48,300\ 481\ 2,700$	6.9 7.4 7.3
							1	2. MOR	GAN CRE	EK NEAR	COLOR	DO CIT	Y					1				
Water year 1947																						
Sept. 10, 1947- Minimum,	28.0			252	113		62	249	1,090	1,000		2.0		3,520	4.54		1,090	890	60		5,020	
Sept. 11-13 Weighted average-	38.0 31.2			26 35	5.4 6.8		26 36	99 104	38 47	13 40		5.2 3.3		171 238	.23 .32		87 116	6 130	39 41		276 387	
Water year 1948 Maximum, May 25, 1948 Minimum, July 6 Weighted average-	7.50 6,043 34.8	13 4.0 		265 14 29	101 .3 4.9	1,4	10 1.4 26	116 32 103	1,170 2 33	2,000 3.0 20		3.5 8.2 3.8		5,230 72 188	6.84 .10 .26	102 1,170 18	1,080 36 92	966 10 8	74 7 38		8,100 100 306	
								23. В	EALS CR	EEK NEAR	WEST	BROOK										
Water year 1959 Maximum, Aug. 18- 20, 27-28, 1959 Minimum, July 2, 12, 1959 Weighted average-	1.34 164 15.9	5.7 10 8.9		253 24 48	488 5.9 29	2,1	90 33 53	118 96 117	2,030 22 138	3,520 37 233		 1.5 2.3		8,440 180 680	11.6 .24 .92	30.5 79.7 29.2	2,640 84 239	2,540 6 143	63 46 58	18 1.5 4.3	12,400 217 1,130	6.5 7.8
Water year 1960 Maximum, May 5-21, 1960- Minimum, Nov. 4, 1959 Weighted average-	5.02 19.0 33.7	6.9 10 9.6		395 28 44	978 6.1 26	3,	520 14 125	231 106 116	3,810 15 117	6,030 16 193	1.3	2.1		14,900 155 585	20.5 .21 .80	202 7.95 53.2	5,010 95 217	4,820 8 122	60 	22 3.7	20,200 258 942	7.4

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2.5 8.1 8.0 7.3 7.1 7.5 7.4 7.3 7.3 7.2 7.6 Hq 7.4 7.8 6,780 219 1,510 9,970 303 830 7,610 393 770 330 Specific conduct-ance (micro-mhos at 25° C) 376 950 8,920 457 1,800 500 322 9,580 10,600 . 10 18 1.6 2.2 16 1.7 2.9 $15 \\ 1.9 \\ 4.8$ 14 .8 4.3 14
 1.4
 2.31.12.4..8 So-dium adsorp-tion ratio 22 29 Per-cent so-dium 44 35 25 69 $64 \\ 51 \\ 51 \\$ 63 23 $^{64}_{44}$ 83 74 Non-carbon-ate 1,870 0 97 1,130 0 189 $^{1,420}_{12}$ 1,34022 173 1,70024 264 840 2035 530 20 Hardness as CaCO, 2 1,300 91 318 127 278 Cal-cium, magne-sium 2,050 75 200 1,7301,890 117 378 620 118 192 904 100 1,440105217 _ 28.5 6.85 23.1 247 25.0 95.1 4 111 Tons per day 39. 55. 49. 803 178 106 $590 \\ 441$ solids .27 9.52 .34 1.50 6.15 .19 1.24 7.30 .29 .60 8.42 9.11 22.23 8.62 .45 29 Dissolved Fons per acre-foot 6 1,080 6,780 170 481 6,100 253 1,100 ,370 215 439 6,190 6,700 6,340 216 569 180 520 141 915 Parts per mil-lion 4 ດົ Bo-Ba-NEAR WESTBROOK--Continued 3.54 3.5 3.2 2.5 2.0 3.7 2.2 3.2 Ni-trate (NO₃) ł SILVER 40 0.3 4.9 4 Fluo-ride (F) NEAR - 00 -1,690 7. 302 $^{20}_{446}$ $^{45}_{188}$ 2,40065 403 020 34 123 120 2,50047.0 32 650 34 148 RIVER (Chlo (Cl) ŝ ¢, ŝ 'n 1,740 21 90 1,520 31 238 1,420 36 81 1,15012 204 570 COLORADO 348 26 1,08029 166 32 118 Sul-fate (SO,) CREEK -i Bicar-bonate (HCO.) 79 97 117 212 91 125 236 114 140 212 126 155 238 129 143 BEALS 241 104 112 128 128 129 24. Po-sium (K) 23. 1,690 33 96 37 123 1,500 48 30 $1,130 \\ 17 \\ 210 \\ 210$ 1,340 35 85 2,030 25 1,930 19 291 1,670 So-dium (Na) 5.3 4.9 5 4 4 σ Mag-ne-sium (Mg) 277 6.: 16 364 6. 20 25. 323 8. 49 330 128 38.6 98 121 192 36 50 220 19 47 150 30 46 33 33 70 310 26 65 379 82 82 82 201 32 Cal-Cal (Fe) 7.0 9.2 12 5.0 8.0 4.6 .9 $^{9.4}_{11}$ 0 Silica (SiO₂) 5.5 7.8 6.4 1.1 101 1,525 11 61.0 12 1.4 362 10.6 5.69 2.73 527 545 35.0 1.0 41.5 6.3 .7 520 37.8 4 Mean dis-charge (cfs) 86. 42. 382 496 Mater year 1961 Maximum, Mar. 11-20,1961 Minimum, Oct.18,19 Weighted average-Water year 1958 Max.mum, 10, 1958 Minimum, 0ct.8-9, 13-14, 1957-----1 Weighted average-Water year 1962 Maximum, 1962-May 1-11, 1962-Minimum, 1-2, 5-8-1 Sept. 1-2, 5-8-1 Weighted average-Mar. 1-31, 1963 Minimum, June 20-Weighted average-Water year 1957 Maximum, 1956 Nov. 3-10, 1956 Minimum, June 1-4, 1957- 7 Weighted average-Water year 1964 Maximum, Apr. 7-30, 1964------Minimum, Sept.12-13 Weighted average-Apr. 1-2, 1965-Minimum, June 12--Weighted average-Date of collection Water year 1963 Maximum. Water year 1965 Maximum, 1964--Water

	Mean	e.11.		Cal-	Mag-	So-	Po-	Bicar-	Sul-	Chlo-	Fluo-	Ni-	Bo-	Dis	solved sol	ids	Hard as Co	ness rCO	Per-	So- dium	Specific conduct-	
Date of collection	dis- charge (cfs)	(SiO ₂)	(Fe)	cium (Ca)	ne- sium (Mg)	dium (Na)	tas- sium (K)	bonate (HCO_)	fate (SO.)	ride (Cl)	ride (F)	trate (NO_)	ron (B)	Parts per mil- lion	Tons per acre- foot	Tons per day	Cal- cium, magne- sium	Non- carbon- ate	cent so- dium	adsorp- tion ratio	ance (micro- mhos at 25° C)	рН
							24.	COLORAD	O RIVER	NEAR S	ILVER-	-Contin	nued									
Water year 1959 Maximum, Apr.21-30, May 1, 1959 Minimum, June 3 Weighted average- Water year 1960	0.64 118 35.7	6.8 14 13		591 49 84	207 11 23	3,90(49 34)	0 9 5	82 154 126	1,760 43 189	6,310 70 534				$12,800 \\ 314 \\ 1,270$	17.5 .43 1.73	22.1 100 122	2,330 166 304	2,260 40 200	78 39 71	35 1.6 8.6	18,200 568 2,120	6.9
Maximum, Apr. 15-25,1960 Minimum, Aug. 20- Weighted average-	.11 335 50.8	$4.3 \\ 14 \\ 12$		572 54 71	165 10 20	3,300 2- 268		$109 \\ 189 \\ 122$	1,740 25 147	5,240 32 415		1.2		11,100 253 1,000	$15.2 \\ .34 \\ 1.36$	3.30 229 137	2,110 176 259	2,020 21 159	77 23 69	31 .8 7.2	$16,200 \\ 414 \\ 1,700$	7.3
Water year 1961 Maximum, May 1-18, 1961- Minimum,	. 02	4.5		693	269	4,51	0	59	2,360	7,180	0.5	2.5		15,000	20.6	.81	2,840	2,790	78	36	22,200	6.4
Weighted average- Water year 1962	118 159	9.6 10		37 51	6.2 12	23 160	5 6	98 124	42 87	32 245		$3.8 \\ 2.7$		204 653	.28 .89	$65.0 \\ 280$	$\frac{118}{176}$	38 75	32 67	$1.0 \\ 5.4$	348 1,130	7.8
Maximum, Apr. 29, 1962 Minimum,Sept.5-9 Weighted average-	93.0 8,356 190	5.0 8.6 9.0		591 32 43	$ \begin{array}{r} 174 \\ 6.2 \\ 11 \end{array} $	4,37(37 201	0 7	139 105 118	1,510 33 75	7,090 45 162		2.0		13,800 228 486	18.9 .31 .66	3,470 5,140	$2,190 \\ 105 \\ 152$	2,080 19 56	43 	 1.6 3.1	20,300 392 827	7.1 7.2 7.3
Water year 1963 Maximum, Apr. 7-17,1963- Minimum,	3.3	6.2		517	190	2,150	16	105	1,550	3,580				8,060	11.0	71.8	2,070	1,990	69	21	11,800	7.2
Sept.9-10,12-13 Weighted average-	9.3 32.3	12 12		36 113	5.9 38	489	6 	93 125	36 275	38 785	.4 	2.8		$\begin{smallmatrix}&203\\1,780\end{smallmatrix}$.28 2.42	5.10 	$\substack{114\\439}$	$38 \\ 336$	33 	1.1 8.8	354 2,980	$7.6 \\ 7.4$
Water year 1964 Maximum, Mar. 1-18, 1964 Minimum,	.7	3.6		378	138	2,230	0	66	1,160	3,610				7,550	10.3		1,510	1,460	76		11,600	6.6
Nov. 19, 1963 Weighted average-	16.0 30.4	7.1 8.9		34 72	6.6 18	24 262	4 2	$107 \\ 126$	27 137	$32 \\ 408$. 4 	$2.0 \\ 3.8$		186 973	.25 1.32		$\frac{112}{255}$	24 151	32 53	$1.0 \\ 5.8$	329 1,690	7.4 7.1
Maximum, Apr. 6-10, 1965 Minimum.	14.6	1.6		466	174	2,290	0	78	1,430	3,760				8,160	11.1		1,880	1,810	73		12,000	7.1
Nov. 18, 1964 Weighted average-	224 125	5.6 9.0		39 61	5.5 13	26 140	6 0	$\frac{108}{147}$	30 87	$\frac{38}{209}$		2.5 3.7		290 595	.27 .81		120 205	31 84	32 52	$\begin{array}{c} 1 . 0 \\ 3 . 6 \end{array}$	364 1,060	7.5 7.4
							2	5. COLO	DRADO RI	IVER AT	ROBERT	LEE										
Water year 1948 Maximum, Feb. 21-25,1948 Minimum,	4.68	7.6		218	57	63(0	129	628	985		0.2		2,730	3.52	33	778	673	64		4,210	
July 6-10 1 Weighted average	1,510 304	21 14		$30 \\ 44$	6.2 9.8	39 104	9 4	111 119	42 80	34 138		2.8 2.8		228 475	.31 .63	7,150	100 150	9 53	46 60		382 796	

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	Hd			ł	8.3	l	7.8	7.9	1 7	7.6	
Specific conduct-	ance (micro- mhos at 25° C)			5.310	297	21.1	6,340	334	1,000 8 700	511	
So-	adsorp- tion ratio										
Per-	so. dium			78	23	ŝ	77	32	7 99 9 99	46	
ness cO.	Non- carbon- ate			435	31	0	608	33	1.500	36 84	
Hard as Co	Cal. cium, magne- sium			554	113	707	722	107	1.560	123 246	
ids	Tons per day			479	2,750	707	887	112 247	151	L,230 182	
solved so	Tons per acre- foot			4.12	25	-	5.05	.79	7.53	1.21	
Dis	Parts per mil- lion			3,030	186		3,710	202 583	5,540	294 888	
Bo:	(B)	neđ									
Ni-	trate (NO.)	Contin		0.5	4.4		· 2	1.8	2.0	3.0	NGER
Fluo-	ride (F)	LEE									BALLI
Chlo.	ride (Cl)	ROBERT		1,460	17 131		1,700	31 188	2,230	66 290	RIVER AT
Sul-	fate (SO.)	IVER AT		372	36 85		568	37 110	1,300	$^{41}_{153}$	LORADO
Bicar-	bonate (HCO.)	ORADO F		146	100		138	91 116	76	197	30. CO
Po.	tas- sium (K)	25. COI		925	16 96		100	23 135	120	49 223	
So-	dium (Na)						Ξ.		1,4		
Mag-	ne- sium (Mg)			45	6.8 10		59	7.3 13	116	18	
Cal-	cium (Ca)			148	34 45		192	31	434	28 69	
	(Fe)										
	SIICA (SiO ₂)			8.0	21 16		23	13	5.1	17	
Mean	dis- charge (cfs)			58.5	5,467 240		88.5	206 157	1.0	75.8	
	Date of collection		Water year 1949	Maximum, Nov. 7-10,1948- Minimum.	June 8, 1949 Weighted average-	Water year 1950 Maximum,	May 5-8, 1950 Minimum, Aug. 2-4	18-22 Weighted average-	Water year 1951 Maximum, Apr. 1-10,1951- Minimum Tuno 3-4	Weighted average-	

	9 9	2 7.4	00 7.4 07.2	0.0	9 7.0	0 7.5	3 7.1	-
_	9	998	10,60 24 250	5,50	93	4,33	832	
	1	2.8 3.1	22 5.9	15	1.3 2.8	9.3	1.7 2.4	
	60	1 23	73 23	73	34 45	09	38 44	_
	1.290	34 77	1,440 11 283	650	62 135	832	34 87	
	1,400	152 194	1,550 97 383	740	155 228	940	161 206	
	39.7	2,080	2,980 242 220			-	11	
	5.71	.52	9.38 .20 1.67	4.37	.74	3.85	.63	
	4,200	384 517	6,900 144 1,230	3,210	268 545	2,830	302 463	
-								
LINGER	1.5	2.8	6 6 1 6 1 6 1 6 1 6 1 6 1 6 1 6 1 6 1 6	1.0	3.0	2.5	3.1	
AT BAL								
RIVER	1,680	98 154	3,200 15 488	1,590	62 171	1,130	58 138	
DLORADO	1,000	99 96	1,140 15 235	396	49 118	634	46 79	
30. CC	124	144 142	129 105 121	110	114	132	155 146	
	974	79 110	1,960 13 300	922	36 107	655	45 88	
	124	11	137 4.2 30	80	11	68	7.7	
	355	54 54	395 32 105	165	44 64	265	52 61	
	7.8	12	5.0 9.2 9.1	3.3	9.9 4.0	11	11 8.0	
	ي. ت	2,005 177	160 622 58.7	167	432 60.1	419	395 198	
	Water year 1962 Maximum, May 1-31, 1962 Minimum,	Sept. 5-301 Weighted average- Water year 1963	Maximum, May 2-5, 1963 Minimum, Aug. 14- Weighted average-	Maximum, 15, 1964- Sept. 15, 1964- Minimum,	Sept. 19-30 Weighted average- Water vear 1965	Maximum, Apr. 27, 1965 Minimum,	Apr. 28-29 Weighted average-	

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	Hq		7.6	7.2	7.3	7.5	8.2	8.1 7.4		1	11	7.8	7.9	8.0	7.7	7.8
Specific conduct-	ance (micro- mhos at 25° C)		556	357 494	636	257 441	584	322 466		2,570	281 537	1,920	315 455	1,890	345 642	2,030 240 478
So-	adsorp- tion ratio		0.4	44	1.3	ч <u>е</u> ;	<u>ت</u> .	4.		1	11	l	11	ł	11	
Per-	cent so- dium		<u>م</u>	15	29	۵	14	14		09	20 37	57	334	67	33 43	67 16 37
0.0	Non- carbon- ate		18	15	13	n o	œ	14		00 00 00	36.8	260	15 30	177	40 8	194 3 30
Hardn as CoC	Cal- cium, magne- sium		282	$164 \\ 241$	242	$120 \\ 210$	284	145 224		522	112 168	397	118 151	292	113 180	308 100 146
	Tons per day		56.7	29.0 42.0	3.93		1			250	728	823	630 954	0.60	220 521	,570 ,580 442
lved solid	Tons per acre- foot	-	0.43	38	.50	.33	.46	.24		2.08 1,	.24	1.50	.37 9,	1.46 1,	.52	1.50 1.19 6
Disso	Parts per mil- lion	-	316	204 281	364	146 246	341	179 267		1,670	176 324	1,100	189 270	1,070	209 380	1,220 137 280
Bo.	n (g)															
.: z	trate (NO.)	ABA	1.8	1.5	£.	3.8	5.0	3.8	SABA	1.5	2.9 2.9	2.8	3.5 4 10	1.8	8.5 6.4	3.0 3.0 2.8
Fluo-	ride (F)	SAN S.	0.3	1.65	ų		¢.	12	R SAN	1		1		ł	11	
Chlo-	cI)	IVER AT	20	14 20	54	3.7 14	20	10	IVER NEA	558	15 64	385	22 48	415	8 8 8 73	463 7.0 56
-lus	fate (SO.)	N SABA B	16	14 16	34	7.6 15	20	11	ORADO RI	354	11	231	16 31	182	21 51	154 9.1 32
Ricar-	bonate (HCO.)	89. SA)	323	$192 \\ 276$	280	140 245	336	172 256	1. COL	169	126 161	167	126 147	141	$128 \\ 170$	139 118 142
Po-	m tas- sium (K)		14	13 15	46	0 4.0	21	11 14	6	356	13 47	242	17 34	276	26 63	286 8.6 39.6
50 50	7 E (2)					б, г					6.9		9.4		6.5	1.2
Ma	(M siu	-	31	51 28	41 34	51 20	71 26	54 2J	-	38 45	35 35 11	08	45.0	86 19	34 49 1, (43 33 43 23 23 23 23 23 23 23 23 23 23 23 23 23
3	C in C	-														
	Iron (Fe)	-														
	Silica (SiO.)		12	11	16	10	13	8.0			11	15	11	17	12	14 15 15
Mean	dis- charge (cfs)		66.5	52.6 52.7	4.0	481 180	135	442 249		302	1,474 832	277	18,880 1,309	379	2,167 508	528 17,800 585
	Date of collection		Water year 1963 Maximum, Dec. 1-31, 1962	Minimum, May 8-16, 1963- Weighted average-	Water year 1964 Maximum Aug. 2-6, 1964-	Minimum, Sept. 20-29 Weighted average-	Water year 1965 Maximum, Nov. 1-19,1964-	Minimum, Nov. 20-23 Weighted average-		Water year 1948 Maximum, Oct. 15-19,1947	Minimum, Sept.11-15,1948 Weighted average-	Water year 1949 Maximum,Sept.10- 13, 17, 1949	Minimum, Apr.21-23, 27-30, 1949 Weighted average-	Water year 1950 Maximum, Aug.2-3, 9, 1950	Minimum, Sept.11, 21-23,25, 27 Weighted average-	Water year 1951 Maximum, 1951 July 9-10, 1951 Minimum, May 26 Weighted average-

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7.5 7.6 8.1 7.8 7.9 7.8 7.6 7.7 8.2 8.2 7.9 8.1 7.9 8.1 7.7 Hd 2,710 1,840,350 380 536 Specific conduct-ance (micro-mhos at 25° C) 1,220 205 ,010 310 1,130 174 361 234 317 481 060 289 419 253 333 820 D. ΩÎ. -So-dium adsorp-tion ratio 4.7 .6 9.5 1.2 5.6 1.3 3.5 1.1 3.5 25 6.6 .6 3.5 .8 5.5 .8 Per-cent so-51 23 23 51 27 40 71 20 34 53 30 60 25 34 48 32 37 57 14 26 64 21 Non-carbon-ate ° [14 1.5 232 $^{10}_{18}$ 279 6 14 232 41 175 29 48 166 220 $^{18}_{28}$ 60 280 Hardness as CaCO. 344 360 91 172 104 Cal-cium, magne-sium 284 83 116 271 118 143 434 78 132 380 114 137 436 332 124 6,4401,850 10,2201,230 543 4,000 504 5,1003,850 376 7,190 330 730 680 070 959 3,980 308 636 661 945 504 274 Fons per day e) 9 Б, solids 1.43 1.11 .31 .43 1.00 1.50 2.07 28 .41 26 86 26 1.62. 14 34 1.50.17 Fons per acre-foot Dissolved 818 227 315 1,0501,100 190 1,5201,100 $149 \\ 204$ 148 304 732 127 184 189 266 633 190 278 102214 $171 \\ 242$ Parts per mil-lion Bo-SABA--Continued 2.0 2.5 3.40 3.8 2.0 2.0 2.0 3°2 4.0 3.4 3.2 2.5 3.5 3.0 2.0 2.2 Ni-trate NO_x) 1.2 ł 0.2 1 11 ||||ł 1 11 ł 11 1 11 ł Fluo-ride (F) 5.5 SAN 2 ы. 19.3 202 25 56 $16 \\ 29$ 238 48 72 470 16 432 680 $^{18}_{43}$ 352 100 12 57 223 Chlo. (Cl) NEAR 7.4 16 4. . 9. COLORADO RIVER 233 142 14730 30 57 17 29 188 21 261 190 11 38 156 20 40 Sul-fate (SO.) Bicar-bonate (HCO_i) 258 127 151 188 94 142 180 127 145 191 120 125 156 110 160 207 125 148 145 102 129 137129 $^{6.0}_{21}$ Po-tas-(K) 91. 7. 264425 245 275 15 132 20 32 226 13 140 149 28 45 137 So-dium (Na) 00 (C 90 N 0 P 03 4 g 12.4 11.6 44 5. 34 in a ຕ່ຳດ 4.9 Mag-ne-sium (Mg) 22 21 40 39 27 11 $^{42}_{42}$ $^{39}_{46}$ 74 $^{38}_{44}$ 108 26 88 38 44 102 36 100 29 86 28 97 Cal-Cal | | | 0.14 1 11 1 11 | || 1 11 1 11 ł ł [](Fe) 00 ŝ 7.8 Silica (SiO.,) 12 12 13 15 8. 10 10 13.8 13 18 19 25,570 246 6,522 593 92. 35,970 651 1,3407,5505247,264906 22,0301,660 2,04677216,0203,3541,800 180 198161 Mean dis-charge (cfs) 3,640 Water year 1959 Maximum, May 21-23, 30-31, 1959----3 Minimum, Jully 21-31 Weighted average-Water year 1954 Maximum, 1954 Mar. 27-31,1954 Minimum, 0ct. 5-12, 1953 Weighted average-Water year 1955 Naximum, Oct. 2-4, 1954-Minimum, Sept.23-25,1955 Weighted average-Water year 1958 Maximum, 1958----Minimum, Oct 14-18, 24-26, 1957---- 2 Weighted average-Water year 1953 Maximun, July 19, 1953--Minimum, Aug. 22-23, 25-27------Weighted average-Water year 1957 Maximum, Aug. 28-31, Sept. 1-5, 1957 Minimum, July 20-31----- 1 Weighted average-Water year 1956 Maximum, Aug. 21, 24-28, 1956----Minimum, Oct. 1-5, 1955-Weighted average-Minimum, Sept. 11-13---- 3 Weighted average-June 3, 1952--collection Water year 1952 Maximum, é Date

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	Hd		7.8	7.7	7.6	7.3	7.2	7.4 7.6	6.9	7.2	7.8 7.8 7.4	7.7 7.7 7.0		8.0	7.1
Specific conduct-	ance (micro- mhos at 25 ⁻ C)		2,310	237 534	1,410	275 625	4,120	365 755	2,320	276 674	2,260 262 475	2,320 334 530		632	462
So.	adsorp- tion ratio		8.4	.5	4.5	1.6	8. 5	2.0	1	1.5	10,00	5.6 .8 1.0		1.8	œ;
Per-	so- dium		69	33	57	36	59	30	ł	21	6	57 26 26		68 3	24
s Q	Non- carbon- ate		194	3 49	164	6 8 <u>6</u>	708	32	402	10 71	212 2 34	288 15 41		20	29
Hardn as CoO	Cal- cium, magne- sium		338	91 178	298	129	842	$124 \\ 215$	576	102 206	355 126 177	432 126 192		196	167
ę	Tons per day		216	13,480 1,070	5,850	2,400 1,030	1,050	4,720	2,560	1,680	 660,				
solved soli	Tons per acre- foot		1.77	.18	11.1	.49	3.32	. 59	1.85	.52	1.70 .20 .36	1.58 .25 .40		0.49	.34
Dis	Parts per mil- lion		1,300	136 316	817	156 357	2,440	209 440	1,360	155 384	1,250 149 267	1,160 184 291		361	252
Bo-	(B)	ued													
-iz	trate (NO.)	Contin	1.5	4.2	2.2	2.8	4.0	67 59 59 59 59 59	.2	1.0	1.0 3.0 2.0	2.2	IN	0.5	1.8
Fluo-	ride (F)	SABA			0.6				ł	ŵ l	111	111	T AUST	0.5	e.
Chlo-	ride (Cl)	SAR SAN	560	12 63	260	11 84	1,000	40 114	490	21 101	550 3.4 45	540 27 54	AUSTIN A	68	44
Sul-	fate (SO.)	RIVER NI	167	10 43	178	6.0 46	500	26 70	289	13 54	150 7.6 31	100 18 34	LAKE	46	27
Bicar-	bonate (HCO.)	LORADO	176	108 158	163	147 170	162	113 168	212	112 165	174 151 174	176 135 184	107.	178	169
Po-	tas- sium (K)	91. CO	354	12	180	6.2 51	564	25 73	271	3.9 59	329 6.0 29	267 20 33		58	3.7
Ś	dium (Na)									13					25
Mag-	sium (Mg)		35	4.0 13	25	9.0 16	8	7.2 19	53	5.4 14	22 3.3 9.9	28 8.8 12		21	14
Cal.	cium (Ca)		78	30	78	37 53	202	38 55	143	32 59	106 45 55	127 36 57		44	44
	lron (Fe)														
	Silica (SiO ₂)		18	12	13	14	11	13	0.6	11 9.8	10 6.8 8.6	6.2 6.4 8.4		14	9.1
Mean	dis- charge (cfs)		204	36,700 1,253	2,650	5,700 1,073	160	8,367 508	696	4,005 446	358 24000 595	1,670 855 1,122			
	Date of collection		Water year 1960 Maximum, Aur.13-14.1960-	Minimum, Oct. 4-8, 1959- Weighted average-	Water year 1961 Maximum, Sept. 6, 1961	Minimum, Oct. 17, 1960 Weighted average-	Water year 1962 Maximum, June 24-27,1962	Minimum, Oct.10-12, 1961 Weighted average-	Water year 1963 Maximum, May 7, 1963	Minimum, Oct. 9-16, 1962 Weighted average-	Water year 1964 Maximum, May 23-24,1964 Manum, Sept.22 Weighted average-	Water year 1965 Maximum, 25, 1965- Sept. 25, 1965- Minimum, Jan. 23-25 Weighted average-		Water year 1965 Maximum, Dec. 1-31, 1964	Minimum, July 1-31, 1965

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	Hq		0	5		11	7.7	7.6	8.0	7.7	8.2	8.0	8.2	7.9	88.2	7.9 8.2
Specific	conduct- ance (micro- mhos at 25° C)		601	450 526	515	447 487	480	456 464	529	462 497	547	470 522	431	379 384	477 389 408	444 428 431
Ś	dium adsorp- tion ratio			11	ł	11			ł		2.1	$1.3 \\ 1.5$	œ	9.7	ىن مۇ مە	1.0
	Per- cent so- dium		ŝ	32.8	33	33	31	32		24	46	38	25	212	15 27 23	30 30 30
# d	Non- carbon- ate		ŝ	1 8 9 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	30	30	29	31 27	40	3 4 5	12	18 26	20	18 18	26 13	20 16 18
Hardn as CaC	Cal- cium, magne-		101	155	176	159 1 62	167	167 160	166	168 167	151	156	166	150	214 146 162	167 149 154
ids	Tons per day		ää	818 1,070	546	647 898	711	1,110	651	404 804	224	637 596	192	1,090	358 1,180 600	203 1,760 867
solved soli	Tons per acre- foot		44	.35	39	.35	.39	.34	.41	.35	.46	.36	.34	.31	.37	98.5 88.5 88.5 88.5 8
Dis	Parts per mil- lion		200	254 300	288	255 274	288	251 270	299	260 282	340	262 293	250	214 225	272 224 235	262 232 243
	-98 (B)															
	Ni- trate (NO,)	IN	a C	1.2	1.0	1.0	1.0	1.8 1.4	ő	$1.2 \\ 1.0$	1.5	1.5 1.5	3.0	3.2	3.5 2.0 1.9	1.2 1.0 1.0
	Fluo- ride (F)	AT AUST		; !	ł		ri	0 <u>0</u>	ņ	N. C.	2			n n	140	ຕູຕູຕຸ
	Chlo- ride (Cl)	RIVER	C Y	45	58	47 50	49	48 48	67	40	70	47 61	35	26 27	28 28 28	42 410 410
	Sul- fate (SO _i)	OLORADO	e r	27 35	32	30	31	30	39	35	38	27 35	20	17	20 16	22 21 22
	Bicar- bonate (HCO,)	114. 0	186	149	179	158 170	169	166 162	154	164 161	170	168 160	178	161 164	230 174 182	180 162 167
	ro- tas- sium (K)		36 36	27 39	40	31 36	34	32 34	88	25 37	29	36 44	25	18 19	17 25 22	30 29 31
	So- dium (Na)															
2	mag- ne- sium (Mg)		16	14	16	15 14	14	14 14	16	16 17	16	13	13	33	17 120 120	11
	Cal- cium (Ca)		5	39 40	44	39 42	44	44 41	40	41 39	34	41 40	45	42 43	58 45 58	445 42 22 22
	Iron (Fe)		0.02		1		ł		.03	11		11		: :	::::	10.11
	Silica (SiO ₂)		a	10	7.2	6.6 6	15	12	12	9.2 11	11	11	12	11	12 12	9.4 7.2 8.6
M	dis- charge (cfs)		1 136	1,193	702	939 1,214	914	1,643 $1,263$	806	576 1,056	244	901 754	284	1893 921	487 1,952 945	2,805 1,322
	Date of collection		Water year 1948 Maximum, Oct 1-21 1947	Winimum, Sept. 1-30,1948 Weighted average-	Water year 1949 Maximum,Jan.1-14, 17-31, 1949	Minimum, Oct. 1-4, 6-31, 1948 Weighted average-	Water year 1950 Maximum, Jan. 1 - Feb. 28, 1950	Minimum, May 1-31 Weighted average-	Water year 1951 Maximum, Sept. 1-30,1951	Minimum, Oct. 1-31, 1950 Weighted average-	Water year 1952 Maximum, Nov. 1-30, 1951	Minimum, Sept.1-30, 1952 Weighted average-	Water year 1953 Maximum, Oct. 1-31, 1952	Minimum, July 1-31, 1953 Weighted average-	Water year 1954 Maximum, Jan.1-31, 1954- Minimum, May 1-31- Weighted average-	Water year 1955 Maximum, Jan.1-31, 1955- Minimum, July 1-31 Weighted average-

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Table 5.--Summary of chemical analyses at daily stations on streams in the Colorado River basin--Continued

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a ł	bH		c t	0	3 8.1 6	3 8.1	4 7.7 9	9 8.5	9 65 	6.7 10	8.2	38 7.6	26 7.6	17 7.6	73 7.2	40 7.5	64 7.2 16 7.3	38 7.6	61 7.6 03 7.2
Specific conduct	ance (micro mhos a 25° C)		ç	9 F	41	43	340	38	33	20	35	45	ň 4	5	ৰ ব	ۍ 	4 10	9	00
So-	adsorp- tion ratio		, -	7.1	.1	1.0	<u>0</u> 6	7.		ł	9.8. 9.8	1.1	<u>ة.</u> ق	.5		1.2	 	1.6	1.1
Per-	cent so- dium		6	55	31	29	18 28	22	203		20	28	29 27	15	23	29	30	36	- 28
neas CO.	Non- carbon- ate			ΩŢ	16	24	16 12	21	12 20	26	17 24	32	17 26	24	31	36	40	57	26.38
Hard as Co	Cal- cium, magne- sium			146	150 146	161	124 124	156	132	175	164 169	181	120	238	186	194	172 186	200	205
ds	Tons per day			1,510	587 841	8,640	2,430 2,660	2,800	3,250 2,540	2,660	881 1,100	935	$^{4}_{2,340}$	702	$^{1,700}_{1,860}$	156	2,090	1,370	239 964
solved soli	Tons per acre- foot			0.34	.31	.35	.25	.32	.26	.39	.30	.39	. 33	.40	.35	.45	.40	.47	.43
Dis	Parts per mil- lion			249	225 234	259	184 201	238	192 216	287	221 249	286	199 246	297	258 276	330	268 293	348	313 338
Ro-	E (8)	hed																	
ž	trate (NO _s)	ontinu		0.7	1.2	4.5	3.2	4.4	4.0 3.8		2.3	÷.	2.0 1.6	5.6	1.0	1.8	.5	.2	1.2
Flac	ride (F)	STINC		0.2	4 ņ	9.	. . 84	.3	ыü	1	5.1	ů	ų ų	5.	ůů.	.2	ůů.	е.	4.0
Chle-	ride (CI)	R AT AU		46	36 40	39	19 30	28	22 24	58	26 34	52	31 41	25	41 41	54	51	87	57 78
1.5	fate (SO.)	DO RIVE		23	22	29	13 17	19	17 19	ł	19 24	28	19 25	24	25	37	33	45	35
Ricer.	bonate (HCO ₁)	COLORA		157	159 158	168	131 137	164	146 163	181	179 177	182	126 160	260	190 193	193	172 177	175	203 182
Po-	tas- sium (K)	114.		33	26 30	31	13 23	20	18 18	1	19 23	33	23 27	19	25 27	37		52	37
3	dium (Na)																34		46
Mag-	ne- sium (Mg)			10	11 9.9	9.2	6.4 7.1	9.9	7.6 9.8	1	13 15	16	9.8 14	19	18	18	18 19	22	20
2	Ca)			42	42	49	38 38	46	40	1	44 43	46	32	64	45	48	39	44	45
	Iron (Fe)								-										
	Silica (SiO ₁)			7.4	8.0 8.2	14	11 9.1	9.4	6 6 6 7 7 7		10 9.6	9.4	12	11	9.2	9.5	11	11	13
Mean	dis- charge (cfs)			2, 243	966 1,331	12,360	4,899 4,900	4,359	6,268 4,353	3 430	1,477 1,631	1,211	9,247 246	875	2,440 2,502	175	2,895 1,414	1,462	283 1,056
	Date of collection		Water year 1956	Oct. 1-31, 1955	Minimum, Apr.1-30, 1956- Weighted average-	Water year 1957 Maximum, June 2-3, 1957-	Minimum, July 1-31, 1957 Weighted average-	Water year 1958 Maximum, Feb. 1-28, 1958	Minimum, Nov.1-30, 1957- Weighted average-	Water year 1959 Maximum, Ann 6 1050	Minimum. Oct.1-31, 1958- Weighted average-	Water year 1960 Maximum,	Minimum, Oct. 8-31, 1959 Weighted average-	Water year 1961 Maximum, Jan.1-31, 1961-	Minimum, May 1-31, 1961- Weighted average-	Water year 1962 Maximum, Mar. 1-31, 1962	Minimum, Nov.1-30, 1961- Weighted average-	Water year 1963 Maximum, Sept. 1-30,1963	Minimum, Oct. 1-31, 1962 Weighted average-

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	Mean dis-		hon	Cal-	Mag-	Ś	Ро.	Bicar-	Sul-	Chlo-	Fluo-	Ni-	Bo-	Dise	wheed sol	ida	Hard as Co	ness CO,	Per-	So-	Specific conduct-	
Date of collection	charge (cfs)	(SiO ₂)	(Fe)	cium (Ca)	sium (Mg)	dium (Na)	sium (K)	benate (HCO _s)	fate (SO.)	ride (CI)	ride (F)	trate (NO,)	ron (B)	Parts per mil- lion	Tons per åcre- foot	Tons per day	Cal- cium, magne- sium	Non- carbon- ate	cent 30- dium	adsorp- tion ratio	ance (micro- mhos at 25° C)	Hq
							114.	COLORA	VDO RIVI	ER AT AU	NILS	Continu	bet									
ater year 1964 Maximum, Dec.1-31, 1963-	43.9	9.7		62	24	4	6	245	47	75	0.4	5 10		389	0.53	46.1	253	52	30		188	7.9
Minimum, July 1-31,1964- Weighted average-	1,568 729	8.1 9.1		42	222	47	4.1	174 183	45 46	828	4.4	0.1		338 351	.46	1,430	196	20 20 20	34	1.5	610	7.5
Maximum, 1965 Maximum, 0ct. 1-31,1964-	132	9.2		48	22	46	3.4	196	45	62	с.	1.0		350	.48	1	210	20 20	32	1,4	628	8.1
Meighted average-	$1,416 \\ 1,475$	7.7 9.8		46 47	14 17	ด้ต้	0 8	181 180	27 32	43	ស៊ីល	2.3		258		11	172 187	24 39	28 28	1.0	475 513	7.5
								133. C	OLORADO) RIVER	AT WHAD	RTON										
ater year 1944 Maximum, Apr. 21-30,1944	1,430			54	17	ž		211	36	51	1	2.0		300	0.41	1,250	205	32	28	1	531	
Sept. 1-10	$\frac{4}{2}, 220$ 2,649			40	12	1.	5	149 175	32	28 41	11	1.5		229 279	.38	2,610	150	27 36	20		352 460	
ater year 1945 Maximum, Mar. 1-10,1945-	3,179			61	16	20		214	34	33	1	2.5		324	.44	2,780	218	42	19		502	ł
Minimum, Jan. 19-241 Weighted average-	14,700 3,766			41 47	7.6 13	, 15	6.3	121 168	22	14 32		1.5		180 255	.35	7,140	134	33.4 33	19		293 413	
ater year 1946 Maximum, Nov. 21-30,1945	2,299	ł		48	19	2,	र्च	174	27	43	ł	1.0		358	.49	2,220	198	37	21	1	474	ł
July 3-6, 1946-1 Weighted average-	11,630 3,535			36 46	7.4	2,1	£0 ₩	118 174	13 27	350		2.1		186 267	.25	5,840 2,550	120 168	10	21		286 427	
ater year 1947 Maximum, Feb. 1-10,1947- Minimum, Aug. 27-31 Weighted average-	3,617 6,792 3,095			50 50 50	18 7.6 15	80.0	9.6	216 114 186	35 31	49 18 38	111	2.0		337 179 280	. 246 .38	3,290 3,280 2,340	231 116 186	33 34 34	23 15 23		546 271 454	111
ater year 1948 Maximum, Apr. 1-10, 1948	976	9.4		48	19	45	~	208	40	64	1	œ		386	. 52	1,020	198	28	35	1	602	1
May 26-31 Weighted average-	3,563 1,246			36 44	7.8 18	21	P-0	131 187	354 35	34	11	1.2		215 310	. 29	2,070 1,040	122 184	15 30	33		365 530	
ater year 1949 Maximum, Apr.1-21, 1949 Minimum, Feb.24-281 Weighted average-	1,419 12,030 1,804	14 8.4 12		3 8 8 3 9 8	14 4.1 11	15	0.0.0	188 104 148	35 27 27	51 26 36		2.2	-	312 144 237	.42 .32	1,200 4,680 1,150	178 87 143	24 21	3333	111	524 240 406	7.8

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	Hd			8.0	7.8	7.6	7.9	8.1	7.5	7.7	8.0	8.0	8.0 7.9	8.1 7.7 	8.2	7.7
Specific conduct-	ance (micro- mhos at 25° C)			531	278 402	555 5	325 513	598	250 474	579 251 353	532	270 406	539 298 431	507 313 435	544	169 331
So- dium	adsorp- tion ratio			ł		1		2.0	.6	1.3 .4	6.	.7	1.1 7.9	1.0 1.0	1.0	4.1.
Per-	cent so- dium			28	23	98	32 32	43	24 34	30 18 23	24	$^{16}_{22}$	28 24	27 24 29	26	17 22
CO,	Non- carbon- ate			34	18 24	96	21 28	0	8 22	24 19 14	26	14 18	10 18	12 18 18	12	3 16
Hard as Co	Cal- cium, magne- sium			198	107 146	174	127 176	175	93 154	211 102 130	203	110	192 104 152	185 114 151	199	66 124
ds	Tons per day			892	1,330	679	715	404	2,970	342 302 766	322	1,380 568	301 1,400 788	358 757 691	235	$^{7}_{3,170}$
solved soli	Tons per acre- foot			0.43	332	45	.28	.48	.37	.45 23	.42	.23	. 25	.39	.42	.15
Dia	Farts per mil- lion			314	160 242	399	206	354	154 263	331 170 211	306	167 239	310 244 244	288 178 246	312	108
Bo-	ron (B)	nued														
-in	trate (NO ₃)	-Contin		1.8	2.2	10	2.5	5.	2.0	3.8 2.2	1.0	2.5	1.5 4.5 1.9	2.4	2	1.0 3.0
Fluo-	ride (F)	RTON		0.2	.;	6		2	6j (j	440	ů.	4.0	4.0.0	1.64		1.4
Chlo-	ride (CI)	R AT WH/		50	18 33	y y	17	64	14 49	54 11 25	41	12	47 20 39	44 24 22	77	9.5
Sul-	fate (SO,)	VDO RIVE		36	21 28	37	34 34	36	30 15	34 19 22	34	21 24	29 18 23	26 231 231	26	11 18
Bicar-	bonate (HCO ₃)	COLOR/		201	108 149	891	130 181	216	104 162	228 101 142	216	118 171	223 116 165	212 118 163	229	77 131
Po-	tar K)	133.						_	6 b	3.5 3.8 8	3.6	3.6	4.8 3.7 8.8	4.8 5.2 2.2	4.9	5.2
Ş.	dium (Na)			60	14	44	3.6	99	16	42 10	30	21	36 27	32 17 29	34	17.0
Mag-	aium (Mg)			15	6.6 10	9	6.7 15	17	5.0 12	18 5.9 8.6	16	5.0	14 5.9 11	11 6.0	14	1.1 9
Cal-	cium (Ca)			55	32 42	43	46 46	42	29 42	55 31 38	55	36 45	54 32 43 23	36 36	57	39
	(Fe)															
1	Silica (SiO ₂)			0.6	13	5	11	11	11	9.0 8.4 11	9.0	17 15	8.6 16 12	8.0 8.0 8.0	8.8	11
Mean	dia- charge (cfs)			1,052	1,637 2,038	767	1,419 892	423	7,153 764	383 659 1,345	390	3,050	360 2,840 1,196	461 1,575 1,041	279	24,360 5,937
	Date of collection		Water year 1950	Mar. 1-31, 1950-	Minimum, Jan. 1-5, 10-16, 20 Weighted average	Water year 1951 Maximum, 91,051	Minimum, June 1-5, 10, 13	Water year 1952 Maximum, Nov. 1-30, 1951-	Minimum, May 25-30, 1952 Weighted average	Water year 1953 Maximum, Maximum, 131, 1952- Minimum, Nov. 1-30- Weighted average-	Water year 1954 Maximum, Feb. 1-28, 1954-	Minimum, Nov. 1-6, 1953 Weighted average	Water year 1955 Maximum, Mar. 1-31, 1955- Minimum, Feb. 6-15- Weighted average	Water year 1956 Maximum, 131, 1956- Jan, 1-31, 1956- Minimum, Feb. 11-17 Weighted average	Water year 1957 Maximum, Dec. 1-25, 1956-	Minimum, Sept. 27-29,1957 Weighted average

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7.9 7.8 7.5 7.4 8.1 8.0 7.7 7.3 7.1 7.4 7.0 7.9 7.8 8.0 7.2 7.9 Hd Specific conduct-ance (micro-mhos at 25° C) 522 204 393 216 372 278 511 645 365 563 280 437 429 199 354 563 491 190 397 606 667 404 584 653 So-dium adsorp-tion ratio 1.0 .5 1.2 1.6.7 1.20.5 с. i 8 4.0 6. 6. 1.7 .9 1.3 1.2 4. 5 12 20 20 22 23 29 35 24 Per-cent dium 35 29 16 23 Non-carbon-ate 30 228 32 22 33 30 53 30 22 22 46 16 31 59 59 50 68 19 45 57 14 33 Hardness as CoCO, Cal-cium, magne-85 147 210 182 82 224 122 168 188 78 148 232 88 148 224 102 187 209 134 199 232 147 203 $^{2,810}_{1,480}$ 2,740 $^{11,990}_{3,490}$ 1,850 9,550 3,250 700 5,020 817 560 869 714 900 580 850 1 11 | || Tons day ъ, ŝ solida 0.35 . 16 .41 .16 .38 .46 .17 .30 .22 . 49 . 28 .54 .45 47 .22 .33 Tons per foot Dissolved Parts per mil-259 211 302 118 231 368 160 303 360 206 323 345 163 243 279 114 231 337 128 223 399 222 328 WHARTON--Continued a : @ Ni-Itrate NO,) 5.0 2.5 4.4 3.0 1.8 2.5 1.5 2.3 1.2 1.8 1.9 5.4 2.0 2.5 1.0 1.2 2.21.9 3.8 3.0 11 18.1 ņ 11 ŝ ł e, e, ł $\left\{ \cdot \right\}$ Ł <u>ຕ</u>ຸ | ຕຸ ю.4 1 11 Files 0 œ 0 AT 27 22. 34 6. (C) 19 (F) 37 27 34. 34. 60 49 $^{28}_{28}$ 98 30 67 72 10 35 RIVER 9 0 27. Sul-fate (SO,) 27 22 22 36 23°8 40 17 26 50 35 35 4545 51 29 42 21 21 32 COLORADO Bicar-bonate (HCO₁) 186 94 153 218 93 159 193 93 153 253 94 154 216 106 190 183 182 200 156 193 204 132 165 133. 3.6 2.6 3.5 3.9 3.3 3.9 ł K) in the P 6.0 | 19 42 17 35 24°31 32 19 58 26 43 42 11 24 5.8 14 25 7.9 18 So-dium (Na) 17 40 3.0 16 4.1 11 4.7 0 0 Ņ 18 6.7 16 Mag-ne-Mg) 14 10.3 23 9. 1.31 21 17. Ξ 5 Cal-Cal 55 29 45 58 26 43 47 26 41 70 30 43 60 30 49 49 52 52 55 51 51 62 41 49 (Fe) 8.4 6.2 10 6.4 8.9 Silica (SiO₃) 7.0 14 1212 12 11 1213 14 14 12 13 13 2,030 11,6301,716 $^{841}_{997}$ 11,270 2,378 1,195 44,130 4,576 620 3,921 620 128 876 820 372 705 836 615 4,560 848 Mean dia-charge (cfa) 37, 6, 27, ຜົດໂ Water year 1961 Maximum, Jan. 16-31, 1961 Minimum, Nov. 1-3, 1960-- 2' Weighted average--Water year 1959 Maximum, Dec.1-31, 1959--Minimum, May 24-25, 1959-Weighted average--Water year 1960 Maximum. Sept.1-30, 1960-Minimum,Jully 262-28 4 Weighted average--Water year 1958 Maximum, Feb. 1-22, 1958-Minimum, 15-19, 1957-Oct.15-19, 1957-Weighted average--Apr. 16-27, 1962 Minimum, Nov. 14-17, 1961 Weighted average--Water year 1965 Maximum, 1965----May 12, 1965----Minimum, Feb. 14-21------Weighted average--May 1-31, 1964--Minimum, Mar. 1-15-----Weighted average--Aug.I-31, 1963--Minimum,Feb.20-25-Weighted average--Date of collection Water year 1962 Maximum, Water year 1963 Maximum. Water year 1964 Maximum,

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Table 6.--Chemical analyses of streams and reservoirs in the Colorado River basin for locations other than daily stations.

						1 [nsəy]		strs pé		on except	wher	e ING	lcateo	Dis	olved so	olids	Hard	ness			Crootfie	
					Max		é	Bi-))	lculat	ed)	as Ca	co3		So-	-uoo	
Date of collection	Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- ctum (Ca)	nag- ne- sium (Mg)	Sodium (Na)	stum (K)	car- bon- ate (HCO ₃)	Sulfate (SO4)	Chloride (Cl)	Fluo- ride (F)	Ni- trate (NO3)	Bo- (B)	Parts per million	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- stum	Non- car- bon- ate	Per- cent so- dium	ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	Hd
							-	LAKE J	. B. TH	OMAS NEAF	3 VINC	ENT										
May 28, 1953 Nov. 11 July 21, 1954 Jan. 3, 1956		10 8.4 8.2 3.6 3.6	0.18 .00 .01 .00 .00	22 16 31 32	0.5 6.4 6.1 0.5 1	70 46 41 53	3.6 3.6 3.6	147 103 104 140 173	61 37 52 52	29 18 22 22 22 22	0.0	4-0 0.0 4-0 4-0 4-0 4-0 5-0 5-0 5-0 5-0 5-0 5-0 5-0 5-0 5-0 5	0.16	277 A203 A189 A225 A225 252	0.31		77 55 61 97 105	00000	65 53 51 51	22	472 313 324 388 429	48.44
Feb. 6, 1957 Mar. 6, 1962 Scpt. 9, 1964 Dec. 21		3.4 2.2 4.0 4.0	. 04	32 34 28 28 28 28	88749 88749 8979 8979 8979 8979 8979 897		60 63 77 77 77	168 176 184 205 194	56 57 53 53 53 53 54 53 56 53 56 53 56 56 56 56 56 56 56 56 56 56 56 56 56	. 28 29 31 31				269 A295 306 308	37 40 42 42		105 115 100 106	00000	56 54 61 62 62	001000 001000	468 494 503 537 537	0.0224
								6.1	DEEP CRE	EK NEAR	DUNN											
May 30, 1964 Junc 2 Sept. 17 May 11, 1965	28.8 1.03 7.80 20	9.0 6.0 17 8.7 8.6		52 36 32 60 24 60	34245	6.6	17 28 15 15 17	170 108 90 126 208	11 21 7.4 14	16 41 9.3 8.5	0.5.5.4.4.	1002		194 190 111 214	0.26 .15 .20 .29		140 108 689 162	- 6000	21 25 15 18	10.6	354 364 166 247 375	6.98 7.7.8 7.7.8 7.7
May 12	3.28 11.1 917	6.3 14 9.0		31 30 61	20.4		- 20 57 -	116 C118 218	25 25 28	10 22 61	4.6.4	ت. 1. ت. ت		146 184 330	.25		87 86 172	090	333 45 72	0.1 1.5 1.9	257 314 591	7.0 8.6
							7	. SUL	PHUR CRF	SEK NEAR	DUNN											
May 7, 1952 Mar. 25, 1953	B.01	2.7		116	39 66		70 105	210 180	340 514	54 78	11	1.0		A771 A1,080	1.05		450 564	278 416	25	-1-0-1	1,090	1- 00
	-					T	3. LAK	E COLO	RADO CIJ	TY NEAR C	OLORA	DO CI	TY									
Nov. 15, 1956 Nov. 6, 1958 May 18, 1960 Dec. 22, 1964		5.3 4.1.2 3.9	0.03	39 38 38 42	9.4 8.4 11 12	56	6.6 37 49 63	167 164 172 185	62 52 64	47 26 37 50	8.0 2 10.	8.01.2		311 A241 A296 328	0.42 .33 .40		136 132 140 154	0000	40000	2112	5588 5588 5588	~ ~ ~ ~ ~
A Residue on eva B Field estimate C Contains the e	poration a	cf 10	°C.	arbona	te (CO ₃									-								

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	Hq				7.7 7.4 7.0 6.7		7.2		7.1		6.8		6.8 6.7 6.9 6.9 6.9		6.3		8.1 8.0 8.0 7.9	
Specific con-	duct- ance micro- nhos at 25°C)		$^{1,890}_{1,710}$		532 515 536 526	1	167		201		8,260		130, 000 196, 000 1153,000 27,500 69,300 153,000		5,090		$^{298}_{1,060}$	
s,	ad- ad- sorp- tion ratio				1.1.0	1	0.5		0.2		17		FOEHDW		10		0.7 3.4 3.9	
	Per- cent so- dium				27 27 35 31	1	22		10		68		76 69 56 76 77		60		22 52 54	(P04)
688 CO ₃	Non- car- bon- ate				73 66 75 80		0	1	0		1,360		7,700 57,700 08,000 08,000 12,000 35,100		1,040	1	5 167 201	phate
Hardn as Ca	Cal- cium, Mag- ne- sium		675 380 580		199 186 170 176		60		81		1,430		47,400 58,400 4,330 12,200 35,500		1,100		113 84 235 276	ppm Phos
olids ed)	Tons per day																	(I); 1.0
olved so	Tons per acre- foot				0.45 .39 .41	-	0.14	_	0.16		8.01				4.41	-	0.26 .18 .86 1.04	Iodine
Dilse (c:	Parts per million		A1,280 620 1,060		A332 285 298 291	-	106		119		5,890		22,300 58,700 185,000		3,240		A192 133 A629 A765	2.7 ppm
	ron (B)	1		CITY		5				SPRIN		NG	4.3					Br);
	Ni- trrate (NO3)	CITY	1.81	DRADO	0.01.02	SPRIM	0.8		0.5	R BIG	3.0	SPRI	49	SING	1.5	ING	4.5 3.0 2.5	nine (
	Fluo- ride (F)	RADO		COLC	14.0	BIG	0.2	ANTON	0.2	VEAI		R BIG	e e	G SPF		SPR.		Bron
	Chloride (Cl)	NEAR COLO	228 94 205	VOIR NEAF	32 36 43 43	REEK NEAF	1.2	NEAR ST!	6.5	I GHWAY 87	1,850	LAKE NEA	$\begin{array}{c} 100,000\\ 122,000\\ 7,040\\ 21,800\\ 69,300\end{array}$	ABOVE BI	1,300	EK AT BIO	14 10 218 275	; 372 ppm
	Sulfate (SO4)	N CREEK	495 235 466	EK RESER	94 94 94	PRINGS C	7.2	LF CREEK	8.0	T U.S. H	,020	DAM SALT	64,400 21,700 80,000 7,720 16,800 52,900	LS CREEK	838	EALS CRE	18 9.4 111 131	ium (Li)
B1-	car- bon- ate (HCO_)	HAMPIO	185 183 127	ON CRE	154 146 116 116	PHUR S	94	7. CA	100	REEK A	89	TURAL	522 884 873 78 218 530	. BEA	64	21. E	132 98 83 91	a Lith
Ğ	sium (K)	(4. C	14 88 37	THAMPI	400 F	SUL	7.2		7.4	ZARD C	59	. NA'	2,330 5,680 50 00 00	20	35		11	16 pp:
	Sodium (Na)		1001	15. (16.	8.8		4.5	18. BUZ	1,500	19	73, 500 63, 000 69, 200 16, 7 16, 7		76			[:(UM) ;
Mag	nie- sium (Mg)		78 33 73		18 16 20 19		2.5		2.7		252		11,200 868 2,540 8,370		213		30 22 23	ganese
	Cal- cium (Ca)		142 98 112		50 395 395 395		20		28		158		535 305 424		88		39 30 58 61	pm Mar
	Iron (Fe)					1		1					0.54					8.
	Silica (SiO ₂)				1.1 .5 2.5 2.5		12		12		4.6		2.211.0		4.5		19 18 12 12	180°(
	Discharge (cfs)														36.1			oration at m Aluminum
ł	uate of collection		uly 8, 1947 ov. 3 ct. 7		ay 17, 1960 ar. 25, 1964 ept. 10 ug. 26, 1965		ay 15, 1965		ay 15, 1965		ay 15, 1965		Mar. 23, 1964 ar. 2, 1965 pr. 6 ay 17 une 21, 1965 ug. 25		ept. 13, 1962		pr. 25, 1957 ay 11 ay 25	A Residue on evap E Contains .95 pp P Density 1.218. A Density 1.280.

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Table 6.--Chemical analyses of streams and reservoirs in the Colorado River basin for locations other than daily stations.

So- Specific con-	atium duct- artsorp- (micro- ium ratio 25°C)		2 198 7.8 3 0.1 214 7.4		3 0.1 222 7.8 5 .1 254 7.8 16 .5 350 7.9 24 .9 374 6.8 24 .9 474 6.8 22 .7 428 7.0		17 245 7.6 20 290 7.6 26 170 7.6 30 255 7.0		25 0.8 377 7.5 94 .7 302 7.1		12 2.6 1,190 6.9 32 1.2 555 6.7 26 .7 316 6.8 49 4.5 2.70 7.3 49 4.5 2.70 7.3 31 2.0 1,380 7.3	28 1.1 568 7.5 42 2.0 725 7.0 43 1.9 643 6.5		28 654 8.0 18 0.5 306 8.2 28 1.4 864 7.2 27 1.3 792 7.7 28 1.4 815 7.1		3 0.0 170 6.6 5 .1 347 6.9 3 .1 316 6.9 3 .1 216 6.9 3 .1 227 6.8	2 .1 318 7.1
88.0	on-Pe ar-ce diste		12 2		10 10 76 2 62 2		23 23 23 23 23 23 23 23 23 23 23 23 23 2		11 2 0 2		198 58 386 344 288 344 288 344 288 288 288 288 288 288 288 288 288 2	108 108 98 98		42 24 0 0 4 2 2 4 2 8 4 2 0 0		იიი-	6
Hardnes as CaC(Cal- Cium, Mag- b b ne- stum		84 103		104 119 148 176 167		107 127 88 102		155 122		334 171 112 567 477	203 196 168		234 136 311 298 300		79 166 154 104	1.50
dids ()	Tons per day																
olved so lculate	Tons per acre- foot		0.18		0.17 .26 .36 .33		0.22 .26 .16		0.30		$\begin{array}{c} 0.87\\ .38\\ .23\\ 1.67\\ 1.09\end{array}$	42 51		0.47 .63 .60 .61		0.13 .28 .25	5
Diss (ca	Parts per million		A121 127		A127 145 190 263 239		162 190 A116 A165		224 167		641 277 168 1,230 798	310 377 334		348 465 439 449		96 204 182 130	100
	Bo- (B)	E			0.06											10.10.10.00	
	N1- trate (NO3)	SRT LF	1.5	KWELL	1.80		1.023		1.0		15 350	2.2	DOVAL	0.0444	KERSL	00.41	10
	Fluo- ride (F)	ROBI	1 0.1	BLACI	38181	KWELL	0000	INTER	0.5	INGER	0 44004	ũũ ị	HRIST		E TANI	0446	
	Chloride (Cl)	RVOIR AT	19 m	DIR NEAR	1. 6. 37 29	EAR BLAC	ထ်ထဲက်မ်	S NEAR W	26 13	AT BALL	235 83 34 540 226	60 131 119	VER AT C	50 9.2 95 103	TER ABOVI	9491	
	Sulfate (SO4)	EEK RESE	24 16	C RESERVO	8.4 35 76 64	CREEK N	322 266 54	MINTERS	22 13	LM CREEK	98 18 150 227	36 57 52	ONCHO RI	22 4.2 16 16	NCHO RIV	4.6 15 10 8.0	20
Bi-	car- bon- ate (HCO ₃)	TAIN CR	92 111	K CREEP	125 133 134 128 128	. OAK	102 140 104 99	. LAKI	176 155	32. El	166 138 102 220 162	192 106 86	SOUTH C	278 171 314 312 306	DDLE CC	93 196 182	100
	Po- tas- (K)	MOUN	1 4.7	. OA	4.5	29	11 11 20	31	n 20		118 118 118 118 118	36 58 58	36.	41 14 56 51 54	IW .	4.1 5.6 5.6	
	Sodium (Na)	26.	1.8	28	3.2						1				37	2 4 1 2 5 2 0	-
	Mag- ne- (Mg)		4.7		4.6 6.9 9.2 13 12 12		4 80 9 9 9 9 9 9 9 9 9		11 9.0		38 16 7.8 51 51	13 19 15		18 21 19 20		6 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7	
	Cal- cium (Ca)	1	26 36		34 36 44 49 47		36 37 26		44 34		71 42 32 107 107	60 47 42		64 90 88 87		28 55 55	200
	(Fe)	1	0.08	1	0.12	-			0.43 .02	1			1				
-	llica 1 SiO ₂) (5.2	1	1.1.68.6		20 20 20 30 30	-	9.0 3.0		4.6 10 7.3 7.8 7.8	5,4.8 8,80 3,88	1	145 445 445 16 16		5.9 13.8 8.8	0.0
	starge Si (cfs) (S			-							1.7 5.2 16.4 1,910	1,730 153 14.8		670 670 111.2 11.1 4.23		318 318 826	0-081
	Date of Di collection		June 19, 1951 Nov. 14, 1956		Oct. 16, 1953 Nov. 15, 1956 Nov. 6, 1956 Apr. 28, 1964 May 11, 1965		Apr. 21, 1950 May 11		Aug. 16, 1963 Nov. 13, 1964		Mar. 11, 1964 Aug. 25 Sept. 1	do May 7		May 9, 1950 Sept. 23, 1964 Apr. 2, 1965 June 7		Aug. 17, 1964 do	AU2 . 19

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Table 6 --- Chemical analyses of streams and reservoirs in the Colorado River basin for locations other than daily stations.

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				-	;		ŝ	Bl-						Dise (c:	solved s	ed)	Hardi as Ca	CO ₃		s'	Specific con-	
te ction	Discharge (cfs)	Silica Ir (SiO ₂) (1	ron Fe)	Cal- tium (Ca)	mag- ne- sium (Mg)	Sodium (Na)	Fo- tas- (K)	car- bon- ate HCO ₃)	Sulfate (SO4)	Chloride (Cl)	Fluo- ride (F)	Ni- trate (NO3)	L I I I I I I I I I I I I I I I I I I I	Parts per nillion	Tons per acre- foot	Tons per day	Cal- clum, Mag- ne- slum	Non- car- bon- ate	Per- cent so- dium	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	рH
							40.	SPRINC	CREEK	ABOVE TA	NKERSL	, Y										
1964	385 7.98 4.14 .06	8.3 8.4 8.7 15.7		44 33 57 57	3.5 3.3 17	8.40 8.70	4.2	147 114 104 189	9.0 9.8 11 68	5.6 8.5 71	0	1.5		152 126 123 374	0.21 .17 .17 .51		124 96 92 212	57 2 4 57 2	35 I 3 9 6	0.1	265 220 212 638	7.1 6.9 6.8 7.4
						41	. DO	VE CREF	ZK SPRIN	GS NEAR	KNICKE	RBOCKE	R									
1949	10.8	29 16		69	20 16	17	12	236 293	24 15	208 21	0.5	6.8 10		A596 316	0.81		254 258	60 18	51	е. 	1,180	7.9
					-		43.	DOVE (CREEK AT	KNICKER	BOCKER	1										
1965	3.04	11		60	20		31	244	24	51	0.6	0.2		318	0.43		232	32	22	0.9	567	7.2
						45	5. TW.	IN BUT'	TES RESE	RVOIR NE	AR SAN	A ANGE	LO U		-							
23, 1964		17 14 2.6 3.6 7.5		47 64 50 49 49	33 22 20 27 19	3.222	288 288 288 288 288 288 288 288 288 288	M158 126 202 174 173	48 32 32 32 47	330 142 195 208 190 126	0 0,0,0,4,4	4 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		737 448 519 617 491 491	1.00 .61 .71 .84 .67		253 188 242 270 208 208	128 84 77 113 65 58	61 51 51 53 53 46 46	000404	1,360 754 1,010 1,130 943 768	8.3 7.0 7.0 7.0
		-					48.	LAKE	NASWORT	THY NEAR	SAN AN	NGELO	-									
1958 1964		7.6 4.8 2.3		133 133 55 55	14 71 250 250	m	39 50 77 36	210 162 207 172	27 318 54 71	56 650 132 165	0 6.4.6.6	3.22		A310 1,610 451 500	0.42 2.19 .61		197 624 244 240	25 491 75 99	30 55 41 47	1.2 6.1 2.7	2,770 835 952	7.9 6.9 7.3 6.9
							49.	SOUTH	CONCHO	RIVER AT	SAN 4	INGELO					_					
, 1947 1952		13		47 68	22		533	226 260	36 57	71 94		0.5		A362 A481	0.49		208 272	23	36	1.76	628 785	8.0
							51.	NORTH	CONCHO I	ALVER AT	STERL.	ING CI	ΤY				-					
1945	1.3 730	20		82 31	38 3.3 3.3	0.9	36 3.9	322 117	45 5.2	61 2.8	0.6	1.5		A484 135	0.66		320 91	56	12	6.0	216	8.1
	- +	1000 s	-	-	-			-				-	-				_				_	

A Residue on evaporation at 180°C. L South Coucho Pool. M Includes the equivalent of 3 ppm carbonate (CO₂). N Middle Concho Pool.

		I	നനനം പ	r-4-0.00 80	500007		6.0 5		4	
-0	DH DH		40-104 40-06	80733	7 4 7 8 8 7		26 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7		30 7	
Specifi con-	duct- ance (micro- mhos a 25°C)		21/ 26(19/	9 6 6 6 6 4 8 6 7 6 6	231 231 231 231 232 232 232 232 232 232		43328		1,26	
-so-	ad- sorp- tion ratio		0.042.0		0004-0		0.2		4.2	
	er- cent so- dium		10 116 117 8	n − 0 0 4	2 13 13 13 13 13 13 13 13 13 13 13 13 13 1		15 18 18 25		53	
1038	in don-		10 P 0 4	600Hu	0 26 2 2 2 0		4000		138 70	
Hardne as CaC	Cal- Caum, Mag- ne- sium		103 122 129 134 88	120 141 128 148 148 166	92 95 100 186 99 112		112 129 132 122 122 163		352 355	
olids ed)	Tons per day									
olved s lculat	Tons per acre- foot		0.17 .21 .23 .16	19 21 20 25 25	.19 		0.19 .22 .22 .22 .31		1.25	
Diss (ca	Parts per million		127 158 170 170	143 157 145 187 198	138 230 134 129		A138 A162 164 229		A920	
	B) (B)			PL	200000		040100	0	- 20	
	N1- trate (NO3)	LSBAD	4.02.0	41 11	⊳4 ωα	INGELO		ANGEL	i	ROCK
	Fluo- ride (F)	R CARI	0	89799	000000	SAN A	0	SAN		DATNT
	Chloride (Cl)	IVER NEAR	3.0 8.5 4.4	1.1 1.1 1.5 1.5 5.2	1089 1089 108 108 108 108 108 108 108 108 108 108	RVOIR AT	4.2 9.6 13 13 13 13 13 13	RIVER AT	300 190	VER NEAR 1
	Sulfate (SO4)	CONCHO R	5.2 9.6 17 10	5.2 2.0 14 12 12 12 12	4,6 15.6 30 3.0 3.0	ELO RESE	4.0 8.4 12 22 22	CONCHO	100 98	MCHO RIV
Bi-	car- bon- ate (HCO ₃)	NORTH	120 142 136 178 102	143 172 159 143	116 132 104 196 118 118	AN ANG	139 158 151 151 191	NORTH	261 348	00
1	Po- tas- (K)	53.	1 4 - 3 4 - 5 7 - 7 7 - 7 7 7 - 7 7	5.1 5.0 6.0 5.8	4.4 11 3.9 13.9 3.7 3.7	6. S	10 110 13 26	57.	80	u
	Sodium (Na)		2.3 6.2 13 13 4.0	1.9 1.09 1.03	4.2 7.1 3.1 1.0	ŝ	4.0			
	Mag- nie- sium (Mg)		4.08 4 4.08 4 4.0	3.1 2.1 6.4	3.0 15 1.8 1.8		5.5 7.2 9.0 13		40	
	Cal- cfum (Ca)		34 37 28	43 44 56 56	32 34 350 42		41-38 380 44-38 380		75	
	Iron (Fe)						0.08			
	Silica (SiO ₂)		10 10 11 7.0	9.0 6.6 5.4 9.1	24 32 16 16 5.5		7.8			
	Discharge (cfs)		234 67.7 1.90 .09 1.43	$255 \\ 75.5 \\ 2.18 \\ 2.18 \\ .68 \\ 115$	3,220 661 13.0 13.0 2,470 68				B3.4	
	Date of collection		May 23, 1963 May 24	Nov. 19 do	Sept. 24 Sept. 25 Sept. 28 Nov. 30 May 16, 1965		Oct. 16, 1953 May. 6, 1958 May. 19, 1960 Sept. 23, 1961 Apr. 5, 1965		Sept. 16, 1947 Sept. 23, 1961	

1,210 717 568 868 824

2.3 1.9 1.6

198 37 127 31 83 31 146 37 127 32

 $374 \\ 231 \\ 188 \\ 266 \\ 284 \\ 284 \end{cases}$

685 0.93 383 .52 312 .41 491 .67 472 .64

0.4 1.2 --- 2.8 --- 2.8 --- 0

196 99 68 132 106

7.7 48 39 71 61

74 46 53 24 44 19 58 29 66 29

Apr. 18, 1946-----Apr. 28, 1948-----July 27------Sept. 21-------Nov. 22------ A Residue on evaporation at 180°C. B Field estimate.

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Table 6.--Chemical analyses of strongs and reservoirs in the Colorado River basin for locations other than daily stations

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Table 6 .---Chemical analyses of streams and reservoirs in the Colorado River basin for locations other than daily stations

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Dato				ç	Mag-		P0-	Bi-					I	Dis	solved s	olids ed)	Hard as C	CO.		-So-	Specific con-	
of collection	Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	ctum (Ca)	nie- sium (Mg)	Sodium (Na)	stum (K)	car- bon- ate HCO ₃)	Sulfate (SO4)	Chloride (Cl)	Fluo- ride (F)	Ni- trate (NO3)	ron (B)	Parts per million	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon-	Per- cent so- dium	ad- ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	Н
						59.	CONCH	O RIVE	R NEAR F	AINT ROCI	KCon	tinue	р									
26, 1949 11, 1964 16 25	6.34 6.34 24 4.43 2,850	20 2.8 13.9		33 155 158 73 50	9.1 84 93 28 11	444	16 58 02 25 25	120 184 154 97 138	15 414 462 80 50	27 345 390 250 38	0.5.4.4	5.1 3.0 5.1 3.0	-	$1,260 \\ 1,260 \\ 1,360 \\ 264 \\ 264$	0.25 1.71 1.85 .81 .36		120 732 776 298 170	21 581 650 218 27	25322 243322 243322	2.5 2.5 8.4 8.4	2,020 2,160 1,120 1,448	6.9 7.0 7.1
. 16	14.5 13.0 4.01 92.1	2.5 2.2 1.1		106 158 136 38	39 84 83 4.9	- 0	55 72 04 19	138 202 176 M116	237 414 380 21	130 360 412 26	4.0000	1.2 8.7 1.2		1,300 1,310 1,310	.87 1.77 1.78 1.78		425 740 681 115	312 574 537 20	22 34 26	1.2 3.8 8.6 8.7	2,150 2,150 2,180 310	7.5 6.9 8.5
						O. MUK	EWATER	CREEK	SUBWATI	CRSHED NO	EN 6 .	AR TR	ICKHAN		-							
5. 6, 1961 1. 8, 1962 13		0.0 2.6 2.8 2.8		22 24 24 22	44 66 1.8 1.8 1.8 1.9 1.8 1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.0	4.6 4.6 7.7 6.1	5.8 7.0 6.1	106 119 118 106 89	0.00440 0.0010	7.0 6.0 11 10 7.5	0	0.0 1.2 8.8 .8		104 115 126 114 98	0.14 .16 .17 .13		8000000 800000000000000000000000000000	00000	15299	00.440	220 234 234 183	7.77 7.24 7.24 7.24
23, 1963 . 23, 1964	$^{2.2}_{13.1}$	4.8 1.6		18 33 24	2.7	3.8 8.8	5.8 10 10	68 115 92	4.0 4.0	6.0 6.3 4.6	<u>9</u> 999	3.20		78 115 104	.11 .16 .14		96 96 98	000	$\begin{smallmatrix}&12\\&7\\2&4\end{smallmatrix}$	<u>0</u> 099	137 217 165	6.2 7.2 6.7
							61.	MUKEW	ATER CRI	EEK AT TR	LCKHAN		1									
t. 18, 1962 t. 18 v 21, 1963 ne 17 t. 23, 1964	0.06 360 145 972 972	7.8 11 9.9 10 9.7 8.0		26 26 28 26 26 28	20.100 20.100 20.400 20.50	3.0 9.0 3.0 0.2 0.0 0 0.0 0 0 0 0 0 0 0 0 0 0 0 0	5.7 5.0 5.0 15 3.2	85 95 125 94 128 128 189	60404 000008	8.0 5.0 14.0 3.2 3.2	000000	1.8 1.2 1.2 2.8 2.8		105 106 131 119 172 172	0.14 114 118 116 23		74 78 104 122 155	1 1 1 1 0	0 6 6 1 6 1 0 1 0 1 0 1 0 1 0 1 0 1 0 1	0 10.00	174 173 222 317 317	6.6 6.6 7.1 7.4
						63.	DEEP	CREEK	SUBWATE	RSHED NO.	1 NE/	AR PLA	CID									
t. 4, 1961		5.9		40	7.2		12	128	19	21	0.4	1.0		170	0.23		129	24	17	0.5	297	7.3
						64.	DEEP	CREEK	SUBWATE	RSHED NO.	2 NE	AR PLA	CID									
t. 4, 1961		3.7		34	3.6	6.0	3.8	102	16	12	0.5	0.8		A142	0.19		100	16	1	0.3	233	7.2
Residue on evap Includes the eq	oration at uivalent c	t 180° of 7 p	C. pm car	bonate	(CO ₃)								-									

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Table 6.--Chemical analyses of streams and reservoirs in the Colorado River basin for locations other than daily stations.

	Нq		7.3	7.0	7.2 6.7		7.3		7.4		6.5		7.3 6.7 6.7 6.7	6.8 6.8 7.1 7.3	7.7 6.9 6.8 6.8 7.0	7.4
pecific con-	duct- ance nicro- thos at 25°C)		331 304	323	364 392 401		348		309		153		223 271 407 379 348	265 458 310 327 236	623 815 719 228 311	261 268
So-S	ad- ad- fion- ratio n		0.7	9.9. 9.	0 O O		0.5	1	0.5		0.2		0.3 8 2 2	1.35	0.1 0.1 0.2 0.2	
	er- so- lium 1		22	22	21 20	1	18		20		12		2 4 4 13 5 4 4 5 5	34 20	41 52 10 19	21 18
88 03	ar- are		37	35	45		40	1	29		4		9 11 10 10	15 24 10 18 7	64 70 85 10 17	ഗര
Hardne as CaC	Cal- Ctum, Mag- h ne- stum		125	122	145 156 158		148		128		63	-	97 112 158 188 174	112 155 130 116	179 184 238 97 126	104 112
olids ed)	Tons per day															
olved so	Tons per acre- foot		0.25	.23	. 28		0.28	_	0.26	URY	0.12	-	0.18 .23 .32 .33	.37 .24	.46 .55 .18 .23	. 22
Dise (c2	Parts per million		181 162	171 179	A208 203 211		208	-	A193	EAR MERC	06		134 A172 A235 A239 A213	A163 A274 178	336 433 406 130 172	161 162
	Bo- ron (B)	ACID				ACID		LACID		EEK) N						
	N1- trate (NO3)	AR PL	1.0	0.8	င ကဲဆ	SAR PI	0.5	EAR PI	0.2	EP CRI	3.0	~	1.2 1.2 .8 .0 .0	60911	2.0	2.2
	Fluo- ride (F)	3 NE	0.4	4.4	444	4 NF	0.4	2 NI	0.3	NG DEI	0.3	DLEMA	0.000000	0.4.0.	ũ⊣4Ω4	ų ų
	Chloride (Cl)	SHED NO.	42 35	37 38	38 440 84	CHED NO.	25	SHED NO.	22	DRY PRON	5.0	K NEAR CO	13 22 34 11 8.7	20 28 28 14	98 158 88 8.2	12 14
	(SO4)	SUBWATER	15	16	18 20 21	SUBWATER	35	SUBWATER	27	NO. 8 (6.8	ED CREEK	10 12 25 8.8	9.2 36 13 19 8.2	46 42 14 16	11
Bi-	bon- ate HCO ₃)	CREEK 3	107	107 114	128 136 137	CREEK	131	CREEK	121	TERSHEL	71	JIM N	107 123 162 216 201	119 160 119 119 102	140 140 187 187 106 133	121 126
5	Po- tas- (K)	DEEP	10 10	9 9	ഗഗാ	DEEP	10	DEEP	4	SUBWA	5.1	71.	3.9 16 4.8 4.8	1 2 2 1 1	57 91 54 14.1	11
	Sodium (Na)	65.				66.	I	67.		P CREEK	4.3		7.2 6.2 4.7		5.4	
;	Mag- ne- sium (Mg)		6.7	6.9	7.3		6.8		6.9	. DEE	1.9		8.44 8.72 9.40 9.64 0.64	404 1.94	12 11 4.8 6.4	3.5 3.0
	Cal- clum (Ca)		39 36	38	50 51 51		48		40	- 69	22		32 50 50 64 8	38 47 1 1 4	55 56 74 31 40	36 40
	Iron (Fe)															
	Silica SiO ₂)		0.0	40	2.2		13		6.7		6.6		8.8 9.9 13.9	86.08 8.08 1.1	1.5 5.2 10 8.0 8.2	20 18 180°G
	ischarge ((cfs)												216 36.2 12.0 2,540 2,080	428 B7.2 117 B5.5 81,470	B30 27.4 8.85 9.34 6.21	1,650 522
	Date of collection		ct. 4, 1961	ec. 6	an. 3, 1962 cb. 2		let. 4, 1961		ct. 4, 1961		let. 13, 1961		ct. 9, 1961 ct. 10 ct. 13 nuly 26, 1962	uty 27 bet 15	hr. 23, 1964 Apr. 24, 24 Jure 15 Sept. 21	May 13, 1965

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Table 6.--Chemical analyses of streams and reservoirs in the Calorado River basin for locations other than daily stations.

lc	DH DH		84 7.2 84 7.2 7.2 7.2 7.2	-	3 6.9		6 6.5 5 7.7 3 7.6		3 6.4		1 6.8		5 6.7		6 6 6 7 3 6 7 7 0 6 7 4 7 4 7 4
Specif. con-	duct- ance (micro mhos a 25°C)		44480 44480		27.		333		54		99		41		44400
\$;	dium ad- Borp- tion ratio		0.5		0.6		0.6 1.5 .6		1.8		2.2		0.8		1.1.0.6.0
	Per- cent so- dium		19 6 20 23 20		22		22 29 21		43		46		24		30 30 30 30 30 30
CO ₃	Non- car- bon- ate		001 96		6		36 126 20		81		64		22		3883388
Hardr as Ca	Cal- cfum, Mag- ne- sfum		108 119 136 108 122		105		123 312 122		138		169		154		146 152 150
solids ted)	Tons per day														
solved s alcula	Tons per acre- foot		0.21 20 25 25 25		0.21		0.24 .68 .24		0.38	1	0.48		0.30		0.32
Dls)	Parts per million		108 A146 181 150 161		152		180 499 176		282		353		219		234 241 235 195
	Bo- (B)]		INWOOL		WOOD	
	Ni- trate (NO3)	LERA	28520		2.0		0.0 2.5	CLYDE	2.0		0.2	BROV	0.8	BROW	0.0 .0
	Fluo- ride (F)	AR VA	0400	ALERA	0.2	EMAN	0.000	NEAR	0.3	RETT	0.2	NEAR	0.3	NEAR	0 0 0 0 0 0 0
	Chloride (Cl)	CRVOIR NE	6.0 25.8 20 21 25.8 20 21 25.8 20 21 25.8	SK NEAR V	19	EK AT COL	33 118 23	AN BAYOU	263 102	DU AT BUF	120	ROAD 2559	43	1 CANAL	58 56 56
	Sulfate (SO4)	REEK RESI	8.8 3.4 8.4 8.8 8.8	ORDS CREI	9.6	ORDS CREI	24 84 17	FORK PECI	106 37	ECAN BAY	35	AT FARM	7.6	WID NO.	20 21 21
Bi-	car- bon- ate (HCO ₃)	ORDS CI	138 146 148 124 138	73. H	117	74. H	106 227 124	NORTH	70	76. P	128	BAYOU	161	COUNTY	132 142 137 126
1	Po- tas- sium (K)	72. H	11 16 14 14		14		16 59 15	75.	48		66	PECAN	22	BROWN	31 31 24 24
	Sodium (Na)		4									77.		78.	
;	Mag- ne- sium (Mg)		40004 608808		3.7		4.4 19 4.8		6.9		7.2		3.5		6.6 6.6 7.4 6.1
	Cal- cium (Ca)		8 8 9 8 9 8 9 8 9 9 9 1		36		42 44		44		56		56		48 50 48 40
	Iron (Fe)		0.09 110 110												
	Silica (SiO ₂)		15 6.4 6.0 2.8		10		8.4 12 11		16 7.5		5.8		6.8		5.9 5.8 6.0 7.0
	Discharge (cfs)				96.8		16.4 .13 250		2.25		2.5		2,320		29.5 11.0 21.2 21.7
	Date of collection		1y 28, 1949 n. 14, 1954 r. 13, 1964 t. 6 1y 19, 1965		y 19, 1965		r. 22, 1964 t. 6 y 19, 1965		y 19, 1965		r. 21, 1964		r. 21, 1964		t. 29, 1963 r. 18, 1964 r. 24 t. 21

A Residue on evaporation at 180°C.

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Table 6.--Chemical analyses of streams and reservoirs in the Colorado River basin for locations other than daily stations.

	Н		6.9		7.1 7.3		7.5		7.7 7.6 7.6 7.6		7.5		6.8 7.2 7.4		 7.3	
pecific con-	duct- ance nicro- nhos at 25°C)		431 548 548 457 562 562		350 433		541		480 509 476 556		314		222 311 333		429 278 1,330 210	
	ad- orp-(1 tion ratio		1.0		0.4 .7		0.3		0.5 .4 .5	1	1.1		0.6 .6 .6		3.6	_
	Per- cent so- 8 dium 1		22 22 23 26 23 26		21		10		13 12 15	1	34		25 20 19		13 15 30 30	_
0°88	Von- car- bon- ate		27 21 31 25 64	-	19	1	10		10 8 3 10		0		000		0 181 8	_
Hardne as CaC	Cal- cium, Mag- ne- sium		151 281 220 157 173 208		160 178		274		234 250 227 258		109		84 130 140		194 123 318 80	_
olids ed)	Tons per day															
solved so	Tons per acre- foot		0.31 54 35 36 36 .42		0.27	-	0.43		0.39 .40 .36 .36		0.25		0.17 .23 .24		0.33 .22 1.09 .18	
Dilse (c	Parts per million		224 400 306 254 264 309		198 241		315		284 293 267 317		A181	-	124 171 178		244 160 A802 130	
	Bo- (B)								0.010.0		~		000		10 20 N O	_
	NI- trate (NO3)	e	0	ITE	8 8 8	E	3.0	e	00000		0.0	BRAD'	000		m N	_
	Fluo- ride (F)	NNWOO	0.4	DTHWA	0	KAVET	0.4	MENAF	0	EDEN	3.0	NEAR	0	RADY		
	Chloride (Cl)	U AT BRO	41 35 36 36 49	NEAR GOL	13 34	FORT MC	17	IVER AT	17 19 18 24	REEK AT	20	SERVOIR	12 14	EEK AT B	13 6.(248 18	
	Sulfate (SO ₄)	CAN BAYO	51 28 28 28 28 28 28 28 28 28 28 28 28 28	N BAYOU	12 16	RINGS AT	11	N SABA R	19 14 13 20	HARDIN C	10	CREEK RE	7.6 8.0 8.8	3RADY CR	9.1 12 146 13	
Bi-	car- bon- ate (HCO ₃)	80. PE(152 293 230 159 181 181	PECA	186 194	32. SP	322	84. SA	274 296 273 302	86.	153	BRADY	105 164 174	88.	237 150 168 88	
ĥ	F0- tas- stum (K)		334 334 334 33	81	21		13		16 116 21 21		26	87.	13 15 15		14 9.9 15	
	Sodium (Na)															
-	mag- sium (Mg)		8.9 15 11 7.2 9.4 18		6.7 6.9		20		20 21 25 25		7.7		4.38		14 9.3 38 3.6	
	clum (Ca)		46 54 54 54		53 60		77		61 58 58 62		31		29 45		55 34 65 26	
	Iron (Fe)															
	Silica (SiO ₂)		6.2 15.2 5.5 4.3		8.3		15		16 12 7.9 15		6.0		6.4 8.2 9.2		19 14 4.6 9.0	180°C
	Discharge (cfs)		0.1 960 .4		477 2,520		12.2		4.02 13.3 29.1 5.14						5,620	ration at
	Date of collection		hpr. 28, 1948 7011y 27 Sept. 21 cov. 23 Peb. 1949 Apr. 21, 1944		Sept. 23, 1964 Nov. 20		Jan. 12, 1965		Nov. 3, 1964 Jan. 12, 1965 Mar. 23		June 27, 1958		Sept. 30, 1964 Jan. 14, 1965 Mar. 24		July 27, 1948 Sept. 21 May 19, 1960 Oct. 9, 1961	A Residue on evapo. P Pield setimate
Table 6 .-- Chemical analyses of streams and reservoirs in the Colorado River basin for locations other than daily stations.

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	Hq		7.6 7.1 6.9	7.0 8.0 7.6		7.6		7.3		6.7 7.9 7.7		7.6		7.6		7.4 7.0 7.4 7.4 7.6	7.8 7.8 7.6	
Specific con- duct- ance (micro- mhos at 25°C)			225 308 546 491	266 365 663		992		518		435 490 455 460		455		449		423 442 404 420 391	447 304 441	
So- S So- S dium ad- sorp-(r tion ratio			0.5 1.1 1.0	6. 1.2		2.3	1	1.5		0	1	0.2		0.2		000040	n 4 n	1
	Per- cent so- dium		25 25 25 25 25 25 25 25 25 25 25 25 25 2	11 19 27		40		37		6 112 12 12 12		n		9		880519 1088	10 14 8	
ess CO ₃	Non- car- bon- ate		18 31 31 31 31 31	10 3 46		28		44		22 19 12		90		œ		401 844 1087	12 16	
Hardn as Ca(Cal- cium, Mag- ne- sium		94 100 199 206 180	115 159 256		314		158		216 240 217 223		231		231		207 215 195 201 194	226 139 215	
ed)	Tons per day																	
solved sc alculation	Tons per acre- foot		0.19 .27 .40 .41	.23		0.75		0.38		0.35 .39 .36 .36		0.35		0.35		0.32 .33 .31 .31 .31 .31 .31	33	
्रध्य	Parts per million		A143 177 291 304 270	154 215 387		548		283		258 262 262 262 262		259		260		235 245 226 241 229	266 172 243	
	Bo- (B)												Hd					
N1- trate (NO3)		inued	0.8 1.8 .5 .2 2.0	3.5	ABA	2.8	RNET	1.2	NOILC	11 4.8 2.2 2.2	EGRAPH	5.2	ELEGR/	5.7	NO	0.8 7.00 8.0	4.5 3.2 2.0	
	Fluo- ride (F)	-Cont:	0.0	ស៊ត	SAN SJ	0.2	AR BUI	0.3	R JUNC	0.000	TELI	0.3	EAR TI	0.3	INCTIC	0.00.00	លុកុកុ	
Chloride (Cl)		F BRADY	19 38 40 340 340	11 17 63	INGS AT 8	159	RVOIR NE/	69	VER NEAD	13 19 19	UER NEAH	6.6	PRINGS NI	8.6	NEAR JI	14 13 14 10	14 14 14	
	Sulfate (SO ₄)	CREEK A1	13 18 28 28 28	13 16 50	ABA SPRI	8.4	IAN RESEI	38	LLANO R1	15 19 15	LLANO R1	6.4	NDRED SI	6.4	NO RIVER	11 11 9,4 12,8	13 12 12	
Bi-	car- bon- ate HCO ₃)	BRADY	92 98 205 236 216	128 190 256	SAN S	348	BUCHAN	140	NORTH	237 269 247 257	SOUTH	273	VEN HU	272	LLA	236 250 228 220 220	262 154 243	
f	Po- tas- (K)	88.	14051	8 4.0	90.	u.	92.	4	93.	4 6 5 2 8	94.	1.4	5. SE	1.5	96	1.5 1.7 3 2.6	20 .9	
	Sodium (Na)			6.7		6		4		9 9 9		5.8	6	6.7		8.4 8.7 5.3 1	8.7 8.7	
;	Mag- ne- sium (Mg)		4.0 9.0 9.2	3.7 5.9 12		35		13		15 20 19		15		15		17 17 17 18 18	18 18 19	
cal- cium (Ca)			31 30 65 67 57	40 54 83		68		42		62 54 58 86 28		68		89		55 50 51 58 58 58 50 58 50 58 50 50 50 50 50 50 50 50 50 50 50 50 50	61 55 55	
	Iron (Fe)																	
	Silica (SiO ₂)		10 10 7.8 9.3 10	8.6 11 8.6		0.6		6.4		15 115 112		13		13		11 12 16 14	14 13 12	180°C
Discharge (cfs)			2,430 186 .28 .21 B.04	2.20 .90 .19		6.62	-			218 47.9 28.9 28.5		17.5		15.7		76.5 70.0 56.6 30.9 533	138 120 106	ation at
	Date of collection		Det. 10, 1961 Det. 11 Npr. 3, 1964 lay 6	Sept. 30 vov. 6 fan. 14, 1965		Jan. 14, 1965		Jan. 13, 1965		Sept. 28, 1964 Vov. 3 Tan. 12, 1965 far. 23		Jan. 12, 1965		Jan. 12, 1965		<pre>dar. 31, 1964 day 5 June 8 fuly 14 sept. 28</pre>	Nov. 3 Jan. 13, 1965 far. 23	A Residue on evapor

	Hq		7.4 8.0 7.3 7.3			7.7 7.7 7.5 7.4	7.12	7.6 7.2 7.2			7.000	
Specific con-	duct- ance micro- nhos at 25°C)		463 524 603 547 598 598 615		355 355 311 362 401	417 417 366 414 403	374 325 325 350	430 475 436 391		443 522 764 779 699	610 597 496 357 404	
s,	ad- sorp- tion ratio		0 1 0 8 8 9 4		00004	00440	က အ 4 က မ	. 9 		1.4	1 8 4 8 4 1	
	So-		221 23 23 23 23 23	1	11 11 12 12	19 14 14	$^{17}_{18}$	15 17 19 18		19 20 30 30 30	20 21 20 20 20	
038	ate	-	26 33 33 33 33 33 33 33 33 33 33 33 33 33		7 16 25 16	16 15 18 18	01151 88	16 14 14		12 22 134 134	52 26 16 26 26	
Hardne as CaC	Cal- cium, Mag- ne- sium	-	218 224 256 225 225 225 226		152 166 150 181 181	178 188 165 188 188	162 172 146 150 154	195 202 185 170		185 214 268 306 258	252 218 165 168	
olids olids	Tons per day											
solved so	Tons per acre- foot		0.36 .40 .39 .39 .45		0.27	32 33 28 29 29	27 31 26 28	334		0.55	44 47 38 38 28 31	
Dis (c:	Parts per million		266 294 332 332 288 332 341		A214 200 184 A197 A224	A237 241 A209 218 212	199 230 191 208 208	234 242 222		A285 A294 402 A387 A404	326 344 281 204 204	
	Bo (B)			1					ΓY			
	Culoride Fluo-Ni- Culo (F) (NO3)		3 5 5 0 0 0 2 8 0 0 2 8 0 0 2 8 0 0 2 8 0 0 2 8 000 8 1000 1000		0.8 1.8 .0 .8	3.00	5.9 4.5 2	19 0 0 0	ON CI	1.2 1.2 .8 1.0	1.0 7.5 1.8	
			484 886 486 486 448 448 448 448 448 448	CANO		0400	04040	040 <u>0</u>	JOHNSO	0.4	0,0,4,0j ⊢	
				VER AT LI	19 16 16 19	26 26 22 22 22 22	22 34 11 18	23 28 27 20	ER NEAR	29 106 102 89	56 84 39 32 32	
	Sulfate (SO4)	AVER CRE	19 28 28 28 28 28 28 28	LLANO RIV	9.5 11 8.2 9.6 13	16 17 15 11	11 12 16 15	16 19 16	ALES RIV	15 35 35 35 35 35 35 35 35 35 35 35 35 35	35 31 20 22 22	
Bi-	car- bon- ate (HCO ₃)	97. BE	234 262 262 256 234 256 256	99. I	177 182 182 191 213	197 207 183 207 205	186 195 165 168 178	219 218 209 196	PEDERN	210 235 236 234 234 234 234 232	244 218 234 182 173	
	Po- tas- sium (K)		14 31 33 33 43		14 9.5 12.6 12.6 12.6	19 18 14	15 11 15 16	16 19 21 17	103.	233 23 23 23 23 23 23 20 23 20 20 20 20 20 20 20 20 20 20 20 20 20	30 268 20 20	
	Sodiu (Na)											
Mag- ne- sium (Mg)			16 33 28 28 29 29 29 29	-	116 116 21 22	255087 55508 55008 555008 55508 55508 55508 55508 55508 55508 55508 55508 5550	20 23 8.2 17 15	23 23 20 20 20 20 20 20 20 20 20 20 20 20 20		20 29 44 29 42	31 32 11 16	e (C0,
	Cal- cium (Ca)		61 44 48 48 48 46 46		33 33 37 38 40	35 33 36 36 36 36 36	32 32 32 32 32 32 32	442 338 55 85 25 85 25 85 25 85 25 85 25 85 25 25 25 25 25 25 25 25 25 25 25 25 25		41 38 38 34 34 35 38 34 36 38 34 38 38 38 38 38 38 38 38 38 38 38 38 38	50 38 41 41	rbonat
	Iron (Fe)											Dm Ca
Date of Discharge Silica collection (cfs)			15 - 7 2 4 8 8 8 8 9 1 1 2 9 4 9 1 2		11 14 114 18 8.2 8.2 8.2	9.0 12 3.5 7.0	8.1 10 15 11 11	6.0 3.3 11 17	1	$ \begin{array}{c} 12 \\ 9.2 \\ 11 \\ 2.4 \\ 9.1 \\ 9.1 \end{array} $	3.2 3.7 114 10 10	180°C
			36.6 9.28 1.42 1.42 1.42		233 216 88 108	114 532 187 97.9 70.5	$^{48.5}_{1,710}$ 1,710 1,020	148 187 304 106			44.4 22.8 .12 276 39.1	ration at ivalent o:
			ept. 29, 1964 ov. 4 ar. 13, 1965 ar. 25 uly 8		pr. 28, 1948 uly 27, ept. 21 ov. 21, 1949 eb. 8, 1949	ay 19, 1960 an 18, 1961 ct. 11 ay 64 ay 62, 1964	unc 8 uly 14 ept. 29	an. 13, 1965 lar. 24 lay 11		pr. 28, 1948 iuly 27 iept. 21 iov. 21	pr. 2, 1964 lay 9 ug. 20 iept. 28	A Residue on evapo • Contains the equ

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Table 6.--Chemical analyses of streams and reservoirs in the Colorado River basin for locations other than daily stations.

	Hq		7.5 7.5 7.1 6.9		7.6		7.5		7.7		6.9		7.0 6.6 7.3 6.9	7.6 6.5 6.7		6.9 6.6 7.2 6.9	7.5 6.5 6.6
pecific con-	duct- ance nicro- thos at 25°C)		758 574 559 254		358		396		434		618		841 371 190 180 222	709 442 300		803 383 204 199 258 258	685 459 292
-os	ad- ad- forp- fion ratio		1.4		0.3			1	0.2		0.4		5.01 1.00 1.00 1.00	1.0		5.0.1.0.0 1	1.1
per- cent so- dium			30 17 55		12	1			2	1	10		26 1347 8	1 23		26 13 12 12	26
ess CO ₃	Non- car- bon-		60 38 0 1 28 0	ĺ	30	1	14		20		29		50 18 28 20 28 20 28 20 28	0 0 0 20	1	85 18 19 11	66 16 4
Hardn as Ca(Cal- cium, Mag- ne- sium		282 246 232 121		166		161		211		301		342 177 107 77 103	296 173 148		310 174 113 87 112	265 200 142
olids ed)	Tons per day																
solved so alculat	Tons per acre- foot	- 	0.56 41 41	-	0.29	-			0.33		0.47		0.72 .29 .17 .14	8 <u>.</u>		0.66 .29 .18 .16	. 55
Dis (c	Parts per million		410 305 301 142		A212	RINGS			239		348		531 210 124 102 123	424		484 211 131 114 1145	404
	Bo- (B)	tinued		TIN		ING SI		ISTIN				NI			NIN		
Ni- trate (NO3)		Con	0.2 1.0 .2	R AUS	0.0	DRIPP		AT AU	4.5	N	7.8	AUST	49 12 1.0 .8 .1.2	17	LSUA 1	30 7.9 2.2 2.2	8.0
	Fluo- ride (F)		0.000	RK NEA	0.2	NEAR	NEAR	0.2	AUSTI	0.2	ET AT	0 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7	÷	SET AT	0 4.0.1.1.1	4	
	Chloride (Cl)	OSNHOL 3	98 41 3.8 3.8	NTGE PAI	20	TY LINE	15	ARTON SPI	12	VINGS AT	22	38TH STRI	61 8.0 9.0 7.0	56 28 4.8	23RD STRI	75 12 2.8 11 12	70 24 5.5
	Sulfate (SO4)	VER NEAF	38 38 31 6.4	AT DOEF	22	AYS COUN		ABOVE B/	18	RTON SPI	22	REK AT 3	45 12 7.0 7.2	40	TEEK AT	61 14 5.4 8.8 11	1 1 23
Bi-	car- bon- ate (HCO ₃)	ALES RI	270 254 232 149	L CREEK	166	EK AT H	216	CREEK	233	11. BA	322	LLER CB	349 194 131 88 116	297 205 174	LLER CH	275 190 137 95 124	242 224 168
ŕ	ro- tas- stum (K)	EDERN	4 6 F 0.0	BUL	0	N CRE		ARTON	0.6	1	ۍ د	WA.	4 2000 2000 2000	90 I I	. WA	33.24	
	Sodium (Na)	103. P	19 19 11 19 19 19 19 19 19 19 19 19 19 19 19 1	106.	-	BARTO		110. B	6.8		-	112	00004 40000	69 T T	113	4 9 . 1 7 . 1 7 . 1	411
Mar	mag- ne- sium (Mg)		43 29 32 5,2		16	109.	16		18		21		1.1.8	7.5		8 2 1 1 2 8 2 4 9 9 4	6 8.1
	Cal- cium (Ca)		42 51 40		40		50		55		86		128 67 40 28 38	106		110 65 32 32 41	06
	Iron (Fe)																
	Silica I (SiO ₂) (1.3 6.5 6.4	-	16				9.3		1		16 5.8 3.3 3.3	¹ ا		14 5.3 4.3 4.6	11
Discharge (cfs)			11.6 74.1 102 .8,600						1.59		75.1		0.30 14.1 830 52.0 26.1	. 49 2.31 43.4		$1.42 \\ 1,380 \\ 76.5 \\ 41.3 \\$.84 42.4 85.8
	Date of collection		an. 16, 1965 teb. 27 une 9		'uly 5, 1960		ot. 17, 1961		Tune 2, 1965		hug. 18, 1965		lot. 5, 1961 lpr. 27, 1962 lune 3 sept. 6	fov. 6		Dct. 5, 1961 hpr. 27, 1962 hune 3 sept. 6	Vov. 6

Table 6 -Chemical analyses of streams and reservoirs in the Colorado River basin for locations other than daily stations.

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A Residue on evaporation at 180°C.

	A Residue on evaporation at 180°C. B Field estimate.														А А									
£.7	664	2.0	τz	30	SLI		₽€.0	842		2·2	2·0	56	36	221	22		L.6	2₫		ττ	09T'E	696T	, Z5,	d₩
					L					YTIC	BAY (AVER NEAR	и одузо	100 · 1	₽£T							4		
7.8	₹I€	G.O	L٦	Þ	621		₽2.0	921		Z.0	S.0	₽Ţ	ΟŢ	εςτ	21		2. <i>T</i>	0₽		21		6961	, 26,	đy
										εE	LE LA	EAT EAG	AL 3J943	.32.	t									
8.7	827 609 927 977 177 287 027	9.0	81	32	202 144 165 206 208 208 202 202		78.0	822A 822A 822A 822A 822A 822A 822A 822A		8-2 0-1 0-1 8'0	8.0	58 38 50 32 33 33 33 34 34	35 31 58 58 30 30 35 35	203 166 175 224 223 223 223 223 223 223 223 223 223	50 22 19 		15 19 1 	19 89 67 		15		1959 1959 1942	, 22 , 3 , 1 , 1 , 1 , 1 , 1 , 1 , 1 , 1 , 1 , 1	AL SC SC SC SC SC SC SC SC SC SC SC SC SC
										LAKE	AJDA3	лек икак и	IN OUAN	сого	τετ									
6'9 9'2	268 292	S .0	8	88 9 T	281 131		0.22	0918		0.0	2.0	18 0'1⁄2	61	176 128	8.4	8.1	6.5	ZÞ	00.00	0.6	018'T 018'5	896T 0961	,18 .1 ,0 9n	э0 пГ
	130° COTOBYDO BIAEB VI COLUMBUS																							
2.7 8.8 9.7	629 282 934	₽`T ₽` 6`	36 12 53	81 8	195 86 514		24. 81. 01.	967 281 808		0. 2. 2.	с. с.	11 25 25	23 5.2 8.0	091 911 540	1 2.5 10 10	с.8 ,	0.0 4.1 7.1	99 28 82		81 0' <i>L</i> 21	02.₽ 02∂,1 80.7	996T	y 3 b, 17, b, 3	De De De
7.0 7.0 7.3 7.5 7.7 7.7	528 819 385 515 024	8.0 7. 2.1 2.1	15 50 50 14 53	₽ ∠ 2 9 ₽ Т	011 222 891 18 281		85.0 81. 81. 81. 81. 81. 81. 81. 81. 81. 81.	SÞI Þ92V 822V 211 082V		0.0 8.5 0.0 2.	5.0 2. 2.	15 97 52 15 34	20 11 20 8.6 8.6	130 520 182 130 130 130	3'J #5 3'5 3'5 3'5	2.8 Z.7	5.5 2.5 2.5 4.2	07 08 29 89		01 22 91 0.0 92	8100 82 92 92 82 82 82 82 82 82 82 82 82 82 82 82 82	6961 6961 6961	r. 17, r. 28, r. 28,	вМ qÅ эU вМ оИ
	JIS9. CUMMINS CREEK NEAR COUNDED																							
9.7	£6¥	ε.τ	33	Lε	128		68.0	282A		0.2.	ε.0	95	34	8Þ1	2.I	22	13	42	00.0	£,6	088'I	Z501	'II əu	ռբ
										CE	и сваи	IVER AT LA	я одаяо	. COI	127									
8 · 7	220	ε.0	οt	82	532		66.0	1-82		ετ	₽.0	75	5∉	293	51	t	9 ° Z	06		7.8	01.0	\$96T	'07 'q	ъэ
									5	TTI I A	тосен	а вуду жа	сев све	MILBAR	.121	[
8 ° L 9 ° L 9 ° L 0 ° 8 1 ° L 1 ° L	964 964 204 204 204 203 203 213	C. C. C. C. C. C.	11 21 21 81 20 50 61	55 42 28 28 9 32	021 922 923 962 081 242		18: 88: 94: 18: 18: 18: 18: 18: 18: 18: 18: 18: 18	552 582 581 582 361 542 354 352		3.8 6.3 8. 0. 0.1 0.1	€. 5. 5. 0.5	21 12 12 21 21 81	74 75 75 77 72 79 79	180 550 588 515 528 528 528 528	91 91 91 67 67 17	[[]]]]	11 91 10 11 11 21 11 21	09 12 06 12 12		2.4 9.6 7.8 7.8 11 7.8 7.8	0.32 2.25 2.67 2.67 2.55	9961 	, 7 , 7 7 9 7 13, 13, 7 13, 13, 7 20 7 21, 13, 7 27, 27, 27, 27, 27, 27, 27, 27, 27, 27	Nai Na Ma Ma Ma Ma Ma Ma
										ЭТ	TAVL	с вегом ре	ON CEEE	INO	120									
Hq	25°C) ance micro- micro- arce arce arce arce arce arce arce arce	ad- tion tion tion tion tion	-T99 Jago -os muib	Non- car- ate ate	sium ne- Mag- Cal- Cal-	голг тэд үяр	Tons per acre- foot	Parts Per noillim	(B) Fon Bo-	(NO ^{\$}) Lefe NI-	(E) Flue Flue	(CI) Chloride	(sOd) (sOd)	car- bon- bon- bor-	-or Po-	(sN) (sN)	-25M -9n muis (3M)	Cal- ctum (Ca)	(Fe) (Fe)	(SIUea) (SUIea	Discharge (sis)	ate of ection	coll	
-noo -oS				CO ³ 688	nbraH As Ca	abilo (bet	Diseolyed solids (bajsluolso)							-ia	-04	<i>'</i> α	-26M	1-0						

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